



# **"Forensic Applications in Continuous Flow IRMS Using the IsoPrime"**

**F. FOUREL**



# Menu

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1/ Introduction to stable isotopes

2/ Dual inlet / Continuous Flow

3/ Example of applications

6/ Conclusions



# Stable Isotope Ratio MS

## ◆ Introduction

Atoms of a particular element having the same number of protons but differing number of neutrons are ISOTOPES.

Stable IRMS is concerned with determining the isotope content in the elements C,N,O,S,H found in light gases (eg: CO<sub>2</sub>, N<sub>2</sub>, N<sub>2</sub>O, H<sub>2</sub>, SO<sub>2</sub>).

IRMS instruments measure the relative abundance of these isotopes to a reference gas (cf: TIMS which determines the absolute isotope ratio).





# Stable Isotope Ratio MS

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When the question triggered by a sample is not

“*How Much?*”

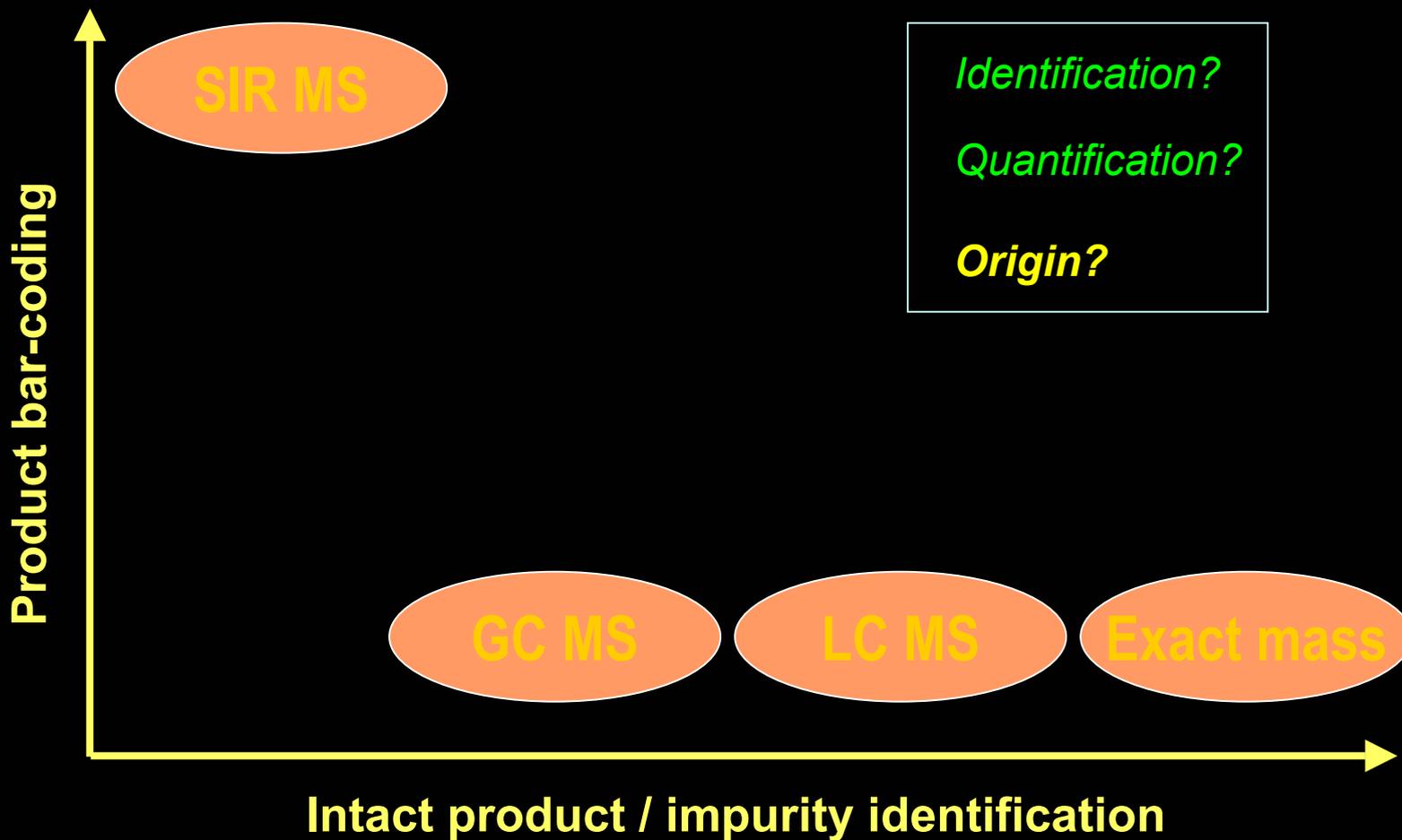
But

“*Where From?*”

Stable Isotope Ratio MS may be the answer...



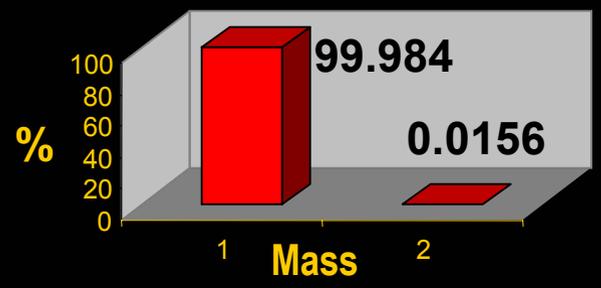
# IRMS for Drug Product Authentication



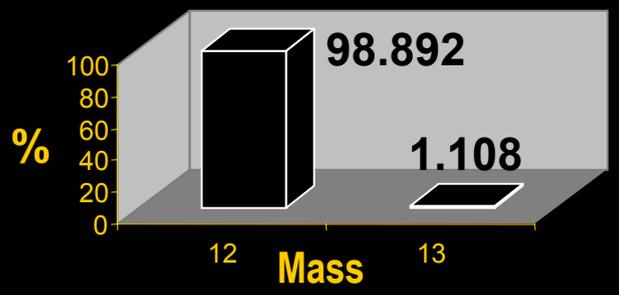


# Stable Isotopes

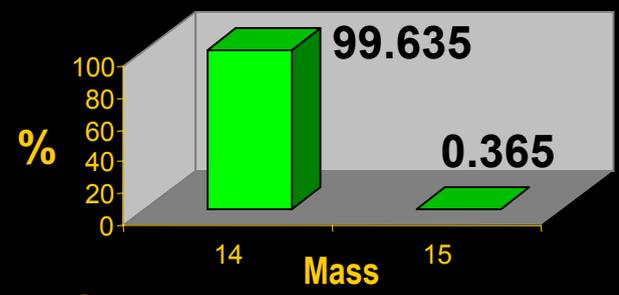
## Hydrogen



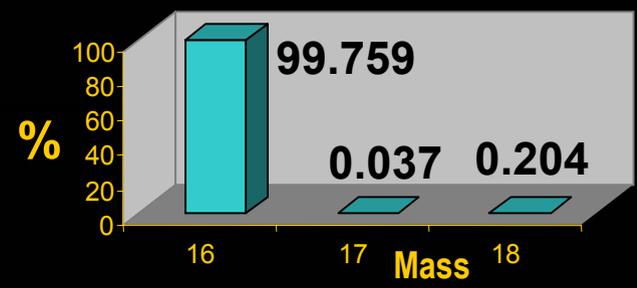
## Carbon



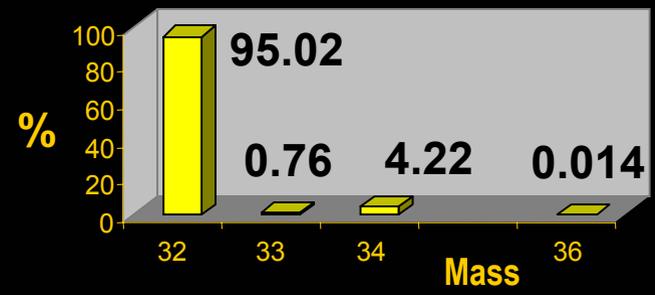
## Nitrogen



## Oxygen



## Sulphur





# Stable Isotope Ratio MS

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The units used to express this relative difference are either Atom% or Delta (per mil, ‰).

$$\text{At\%} = (\text{No of minor Atoms} / \text{No of major atoms}) \times 100$$

$$\text{Delta} = \frac{(\text{Sample Ratio} - \text{Ref. Ratio}) \times 1000}{\text{Ref Ratio}}$$



# Isotopic variations

$$\Delta \text{‰} = \frac{(\text{Sample Ratio} - \text{Ref. Ratio}) \times 1000}{\text{Ref Ratio}}$$

$$^{13}\text{C}/^{12}\text{C}_{\text{PDB}} = 0.011237 \quad \text{International reference}$$

## Examples

$$^{13}\text{C}/^{12}\text{C}_{\text{SAMPLE}} = 0.011248 \quad +1\text{‰ variation vs PDB}$$

$$^{13}\text{C}/^{12}\text{C}_{\text{SAMPLE}} = 0.011226 \quad -1\text{‰ variation vs PDB}$$



# Applications

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## ◆ Natural Abundance

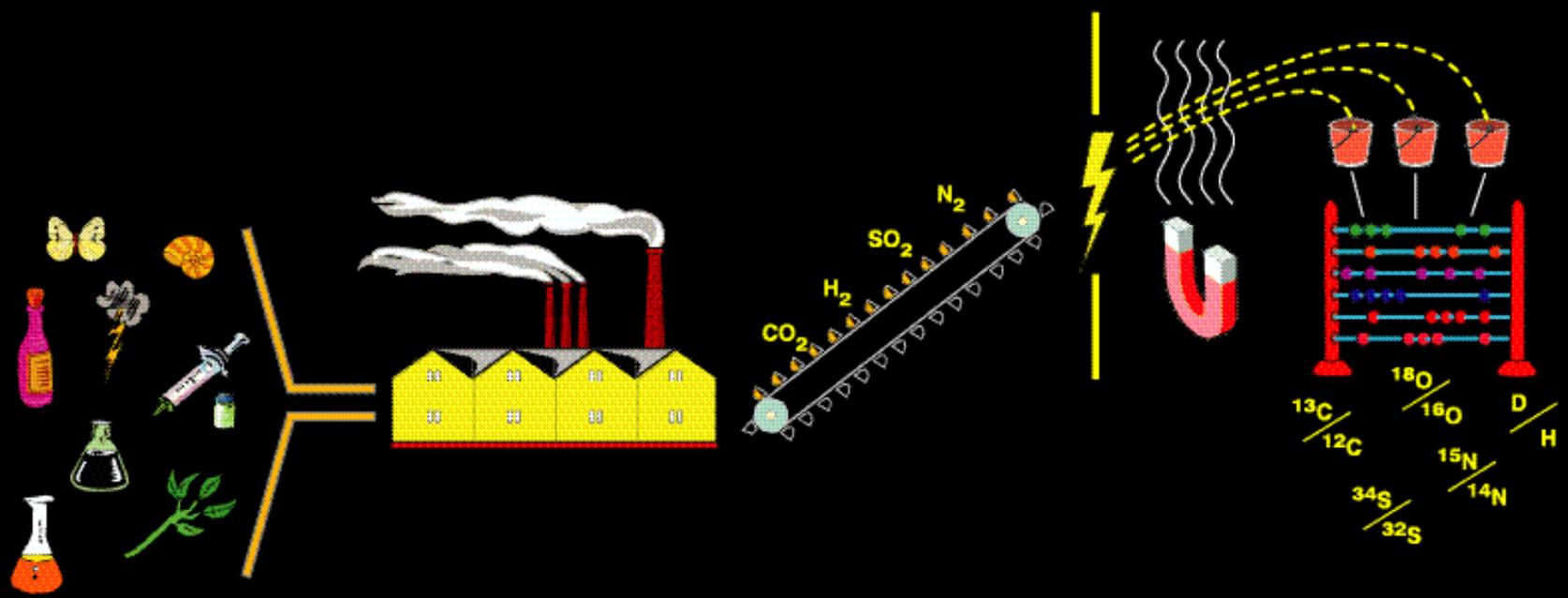
**Natural variations in the isotopic content of substances occur due to the different physico-chemical properties of isotopes**

## ◆ Tracer Levels

**A chemical containing a 'high' dose of a minor isotope is input into a system to enable a process to be followed. (Note: many agricultural and biomedical applications use enriched tracers)**



# What is Stable Isotope Ratio MS?





# Evolution Of Stable Isotope Analyses

## TECHNIQUES

Offline prep - Single Inlet

Water, Carbonates...

Offline prep - Dual Inlet

D/H,  $^{18}\text{O}$ ,  $^{13}\text{C}$ ,  $^{15}\text{N}$ ,  $^{34}\text{S}$

Online - Dual Inlet

Pure compounds

EA - Continuous Flow

EA -  $^{13}\text{C}$ ,  $^{15}\text{N}$

EA -  $^{18}\text{O}$ ,  $^{34}\text{S}$ , D

GC - Continuous flow

Organic mixtures

GC -  $^{13}\text{C}$

GC -  $^{15}\text{N}$

GC - H/D

TraceGas

Atmospheric gases

$^{13}\text{C}$ ,  $^{15}\text{N}$ ,  $^{18}\text{O}$

## APPLICATIONS

Food Adulteration

Soil sciences

Environment

Geology

Medical

Hydrology

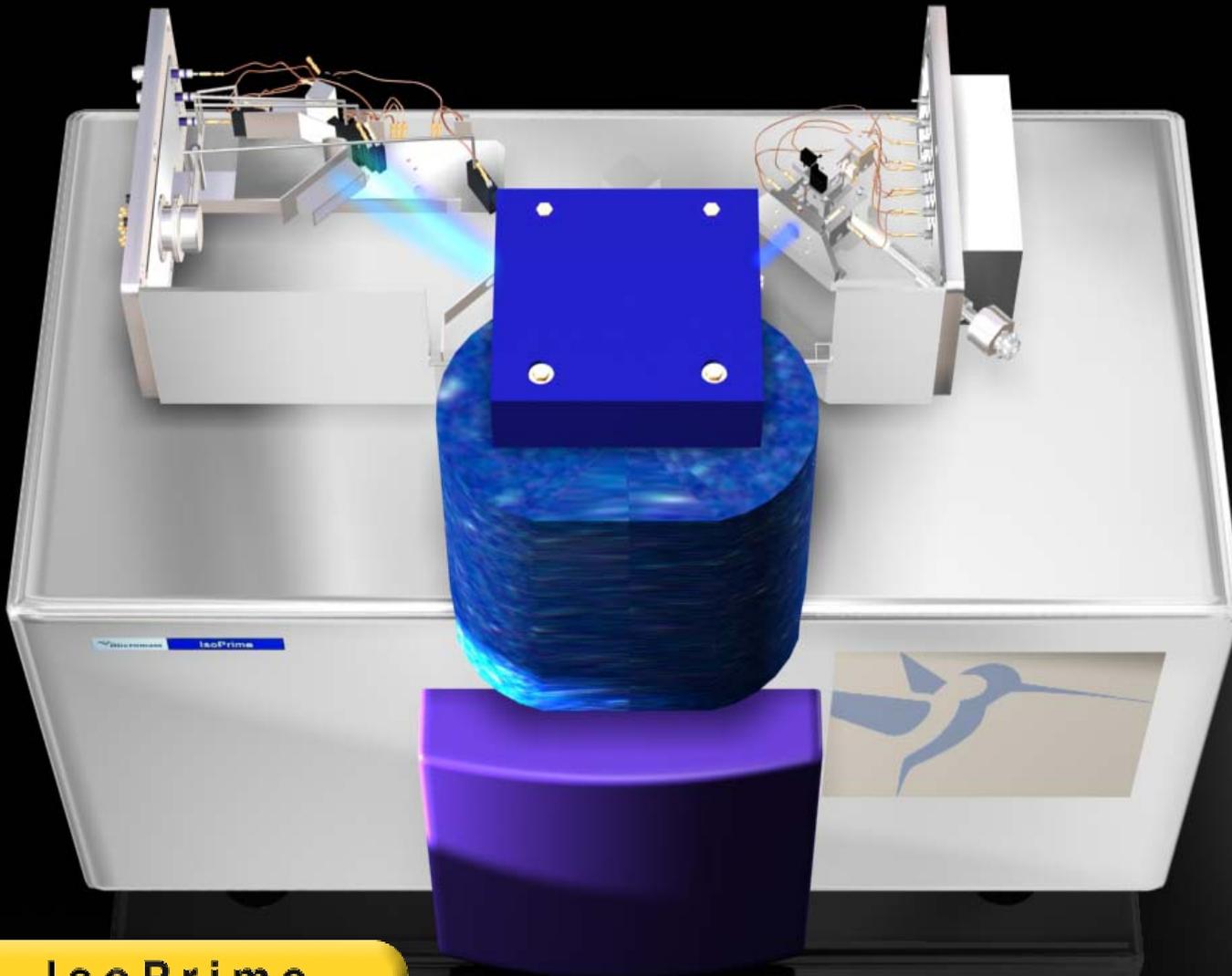


# Key dates in development

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1983	Preston and Owens	Interfaced EA to IRMS	C&N
1986	Hall	GC-C&N	
1987	Pichelmayer	Measured organic sulphur	
1994	Giesemann	EA Inorganic sulphur	
1995	Brockwell	GC-H	
1997	Kornexl	EA Organic oxygen	
1998	Merren	Hydrogen ESF	

# Isotope Ratio Mass Spectrometer

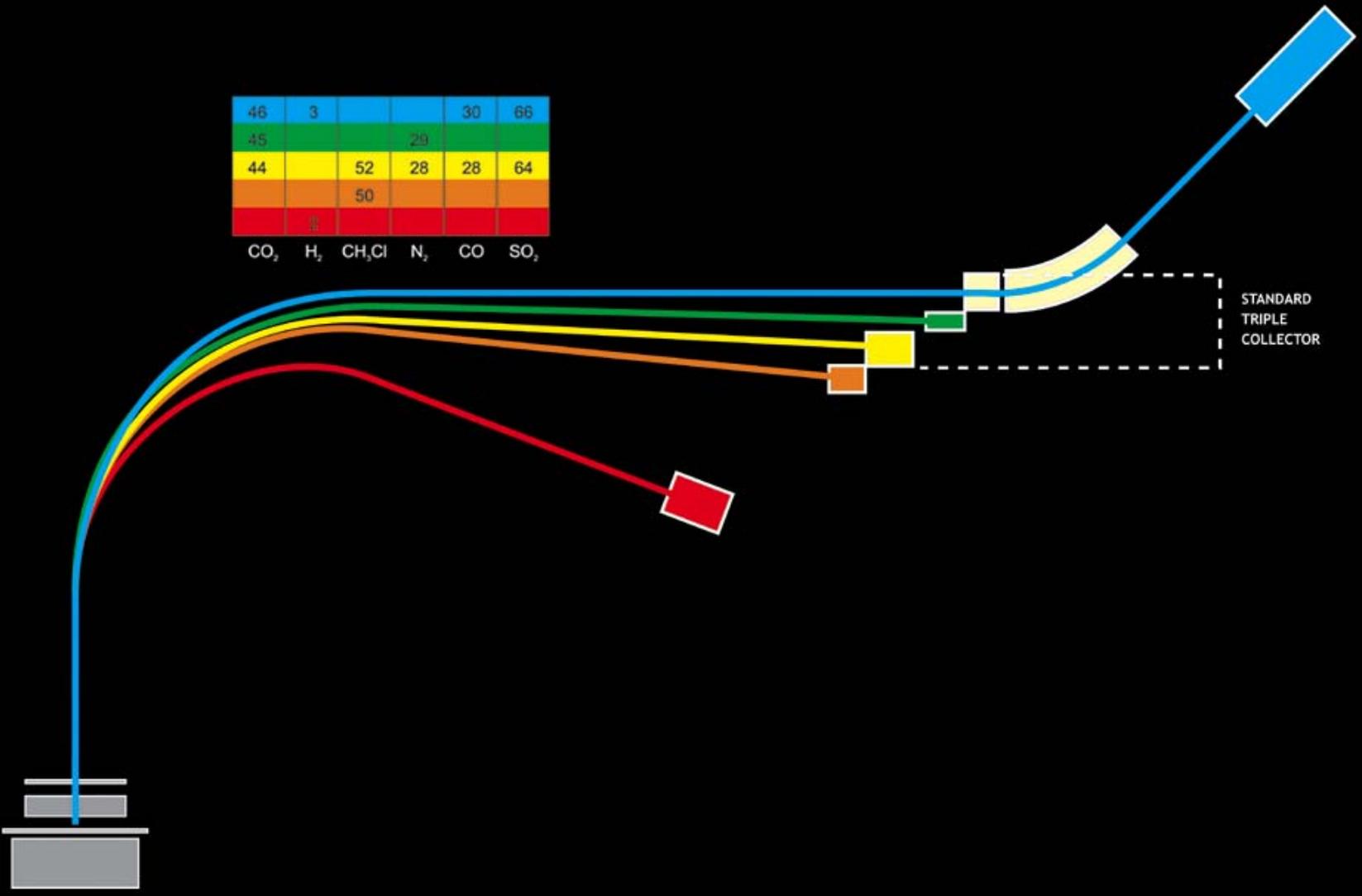


**IsoPrime**



# Collector configurations

46	3			30	66
45			29		
44		52	28	28	64
		50			
CO <sub>2</sub>	H <sub>2</sub>	CH <sub>3</sub> Cl	N <sub>2</sub>	CO	SO <sub>2</sub>





# Menu

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1/ Introduction to stable isotopes

2/ Dual inlet / Continuous Flow

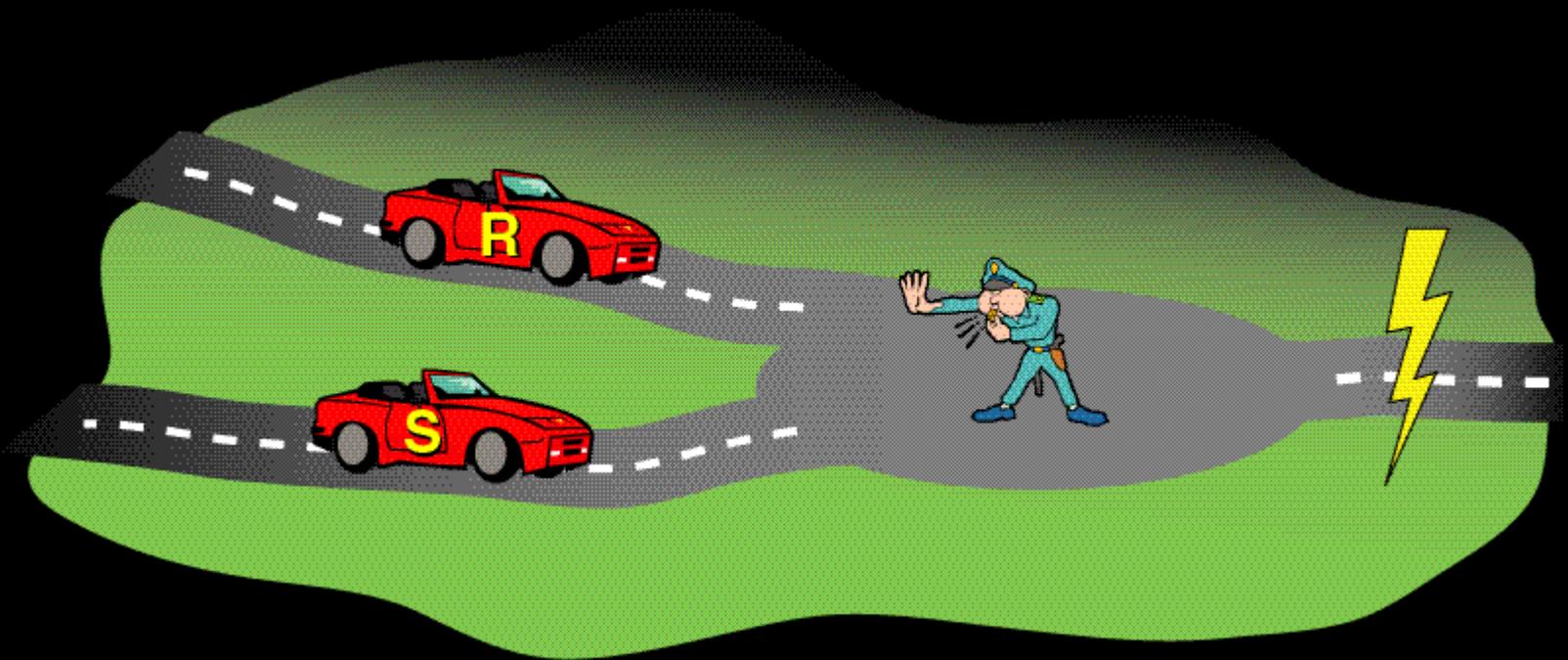
3/ Example of applications

6/ Conclusions



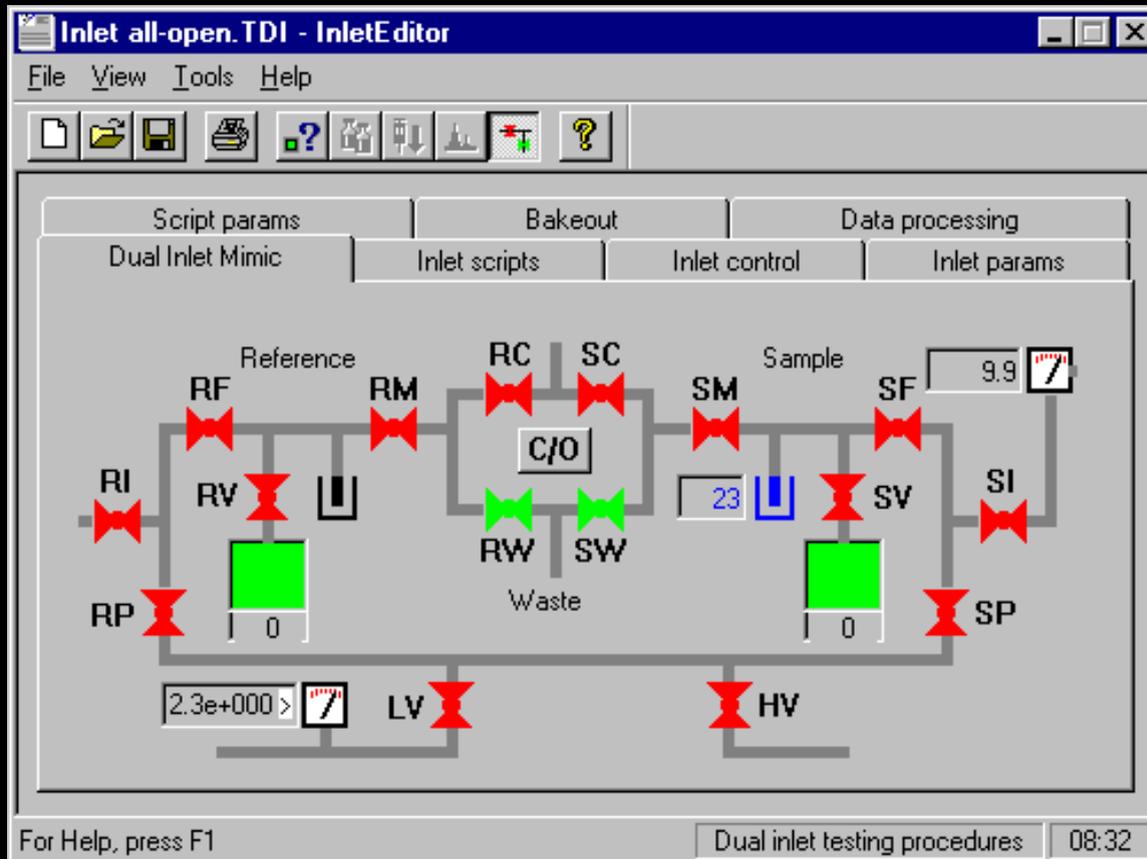
# Dual Inlet

Direct comparison of pure sample gas to calibrated reference gas





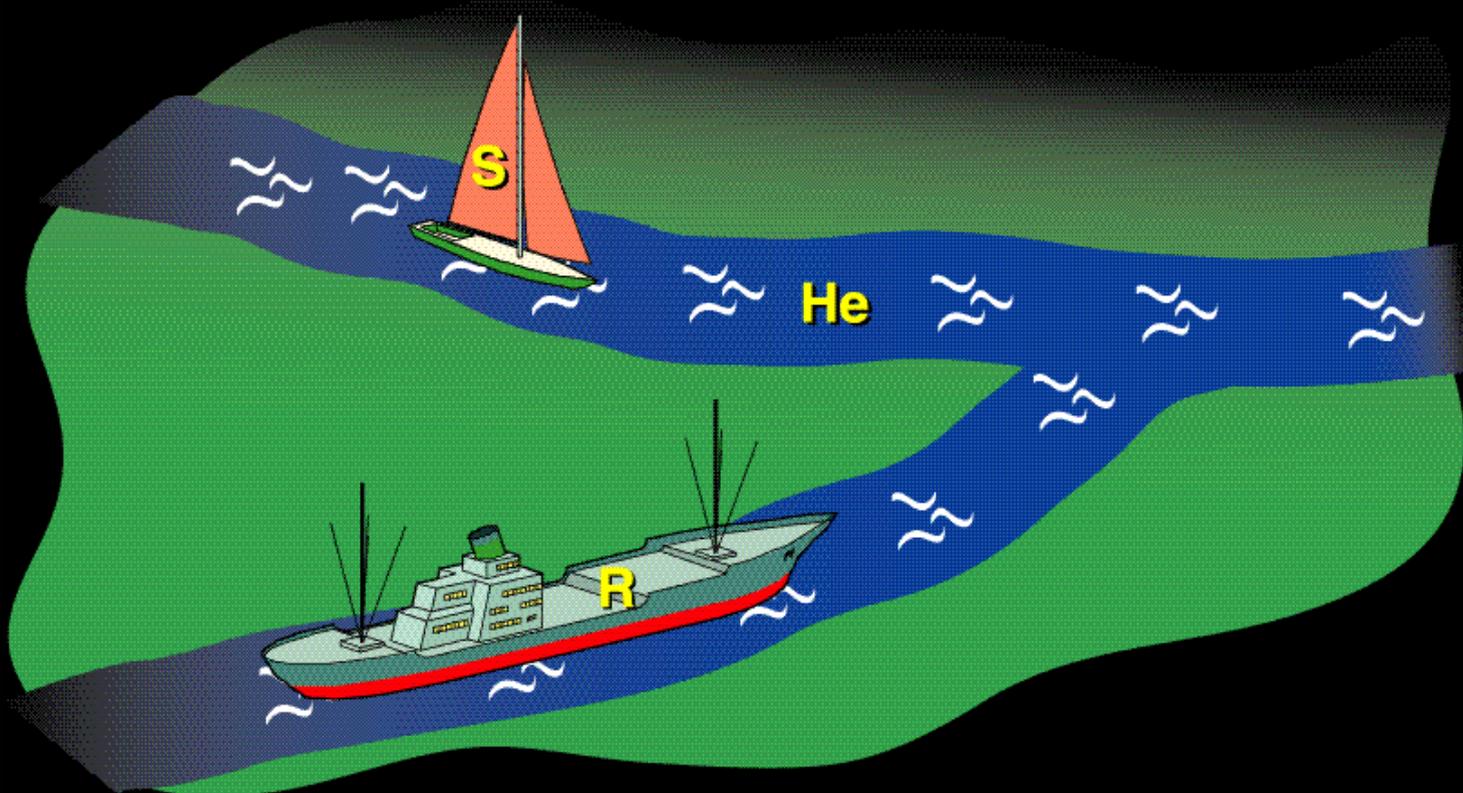
# Dual Inlet Mode





# Continuous Flow

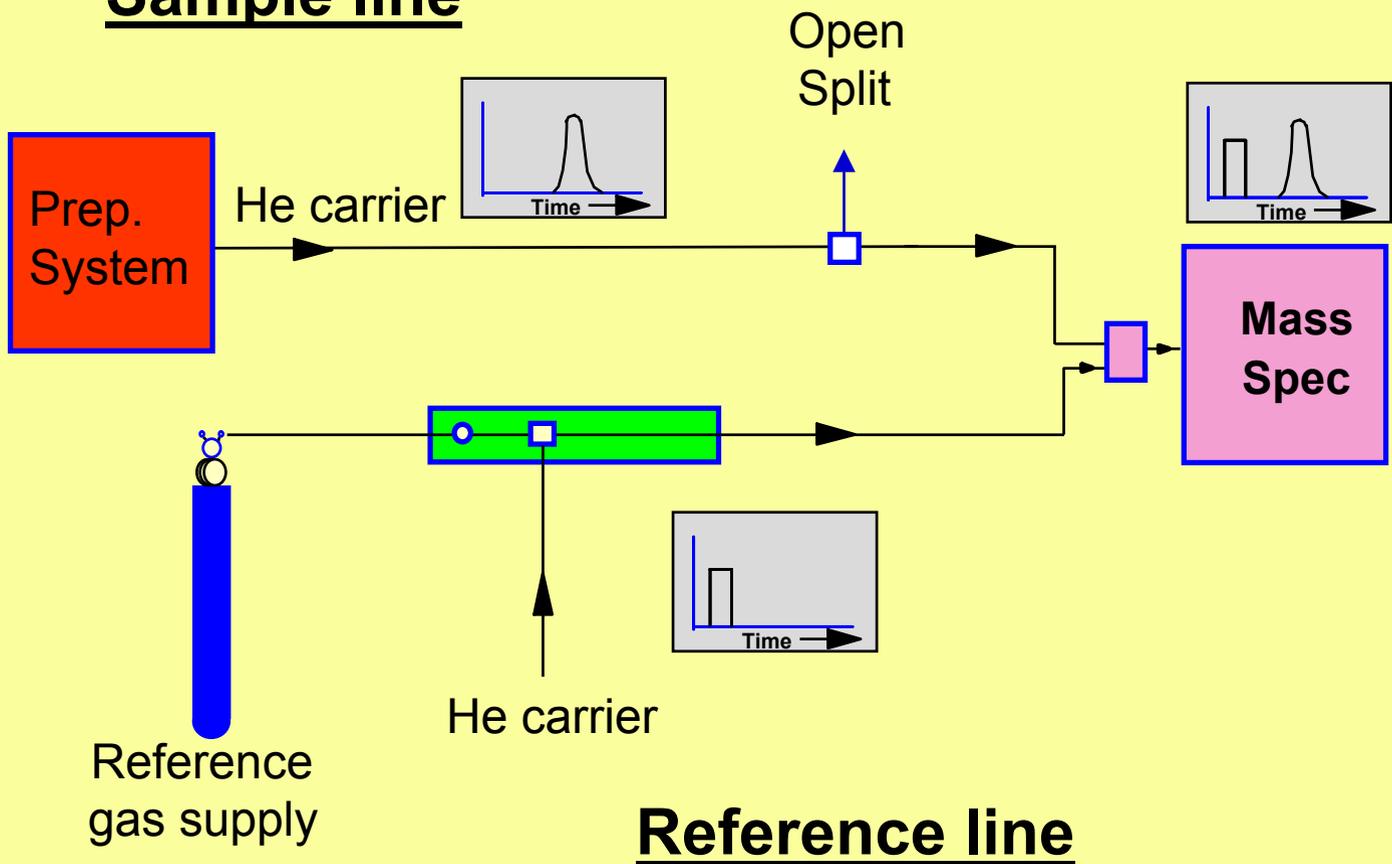
**Sample as a transient signal**  
**Reference gas injected as a gas pulse**





# Continuous Flow

## Sample line





# Continuous Flow

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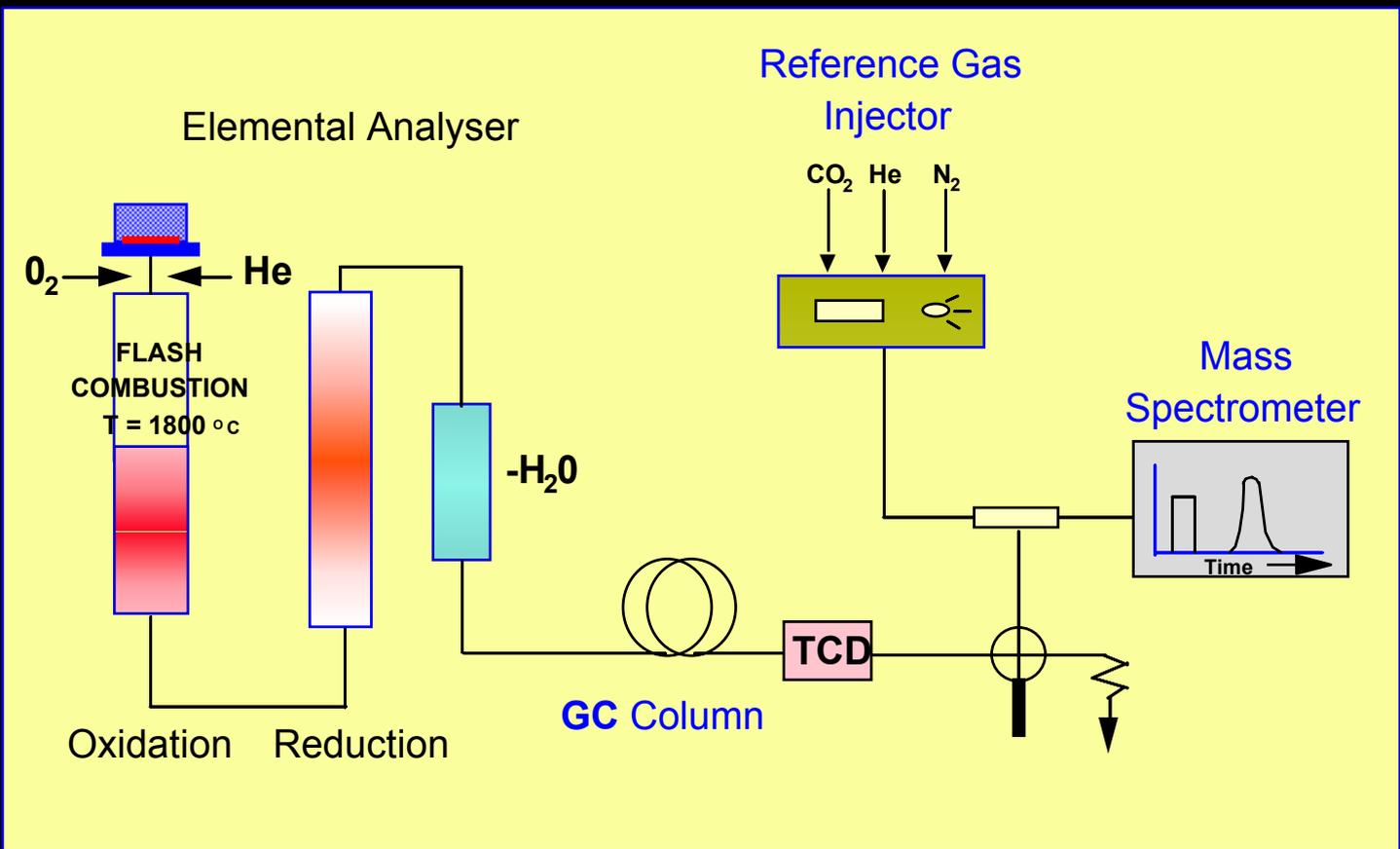
**The advantages of the Continuous Flow (CF) technique are**

- Simpler systems**
- High throughput**

**Continuous Flow techniques offer an alternative, which is faster, simpler and more cost-effective than classical dual inlet preparation systems**

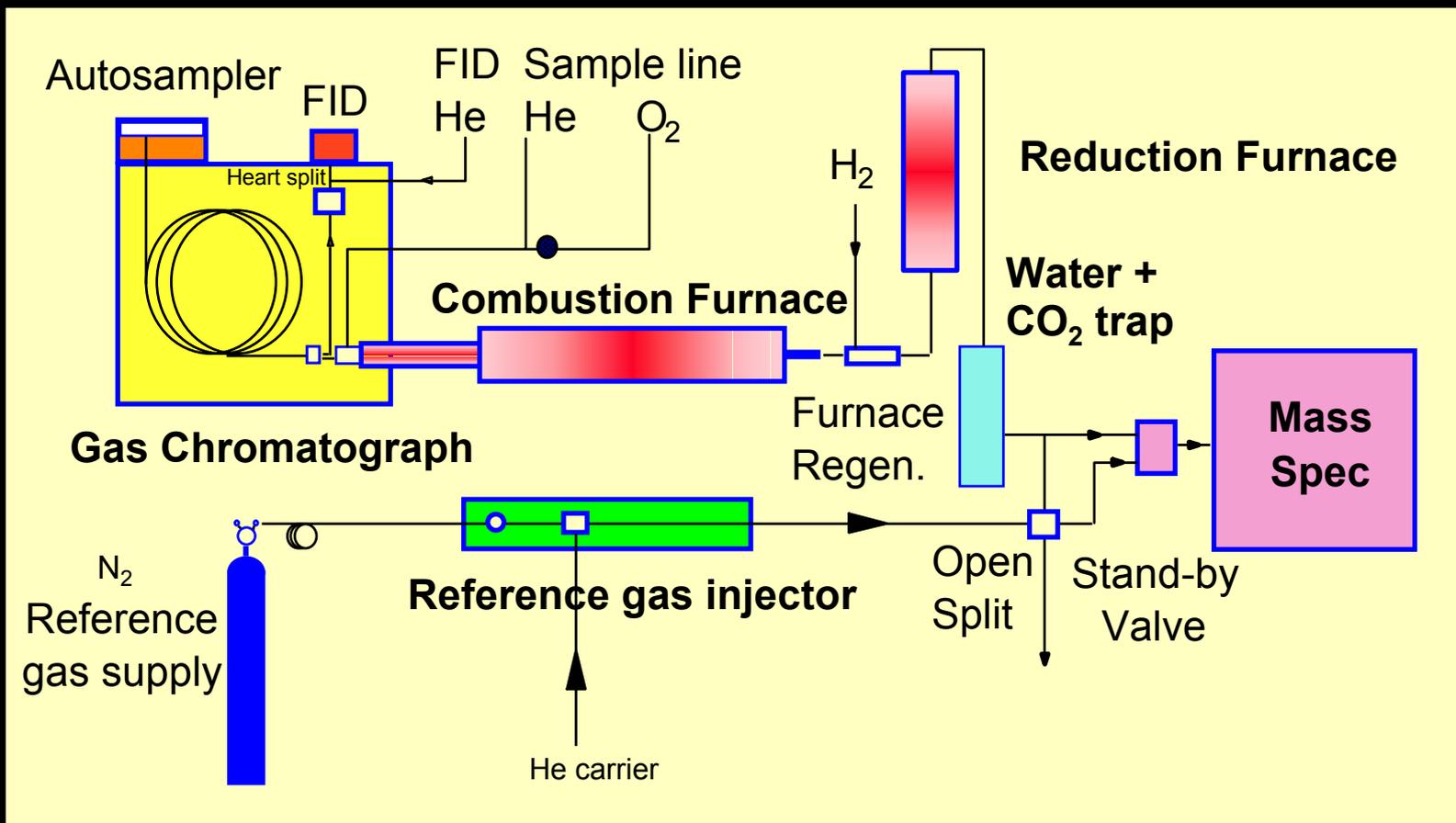


# Elemental Analyser





# GC - Combustion ( $^{15}\text{N}$ )





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# Food Adulteration





## Isotopic ratio

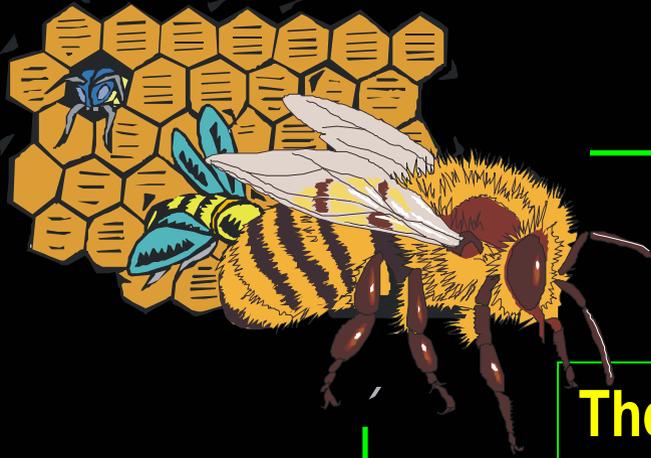
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### Stable Isotope Ratio Analysis (SIRA) by Isotope Ratio Mass Spectrometry (IRMS)

- ◆  $^{13}\text{C}$ : Plants forming sugar through the C3 pathway (beets and most fruits) do not incorporate  $^{13}\text{C}$  in the same way as plants using the C4 pathway (cane, corn), e.g. addition of corn or cane sugar to fruit products.



# Detection of adulterated honey using EA-IRMS



Pure honey

The principle of adulteration detection is to compare characteristics of a product with those of a similar product of known origin

High fructose corn syrup

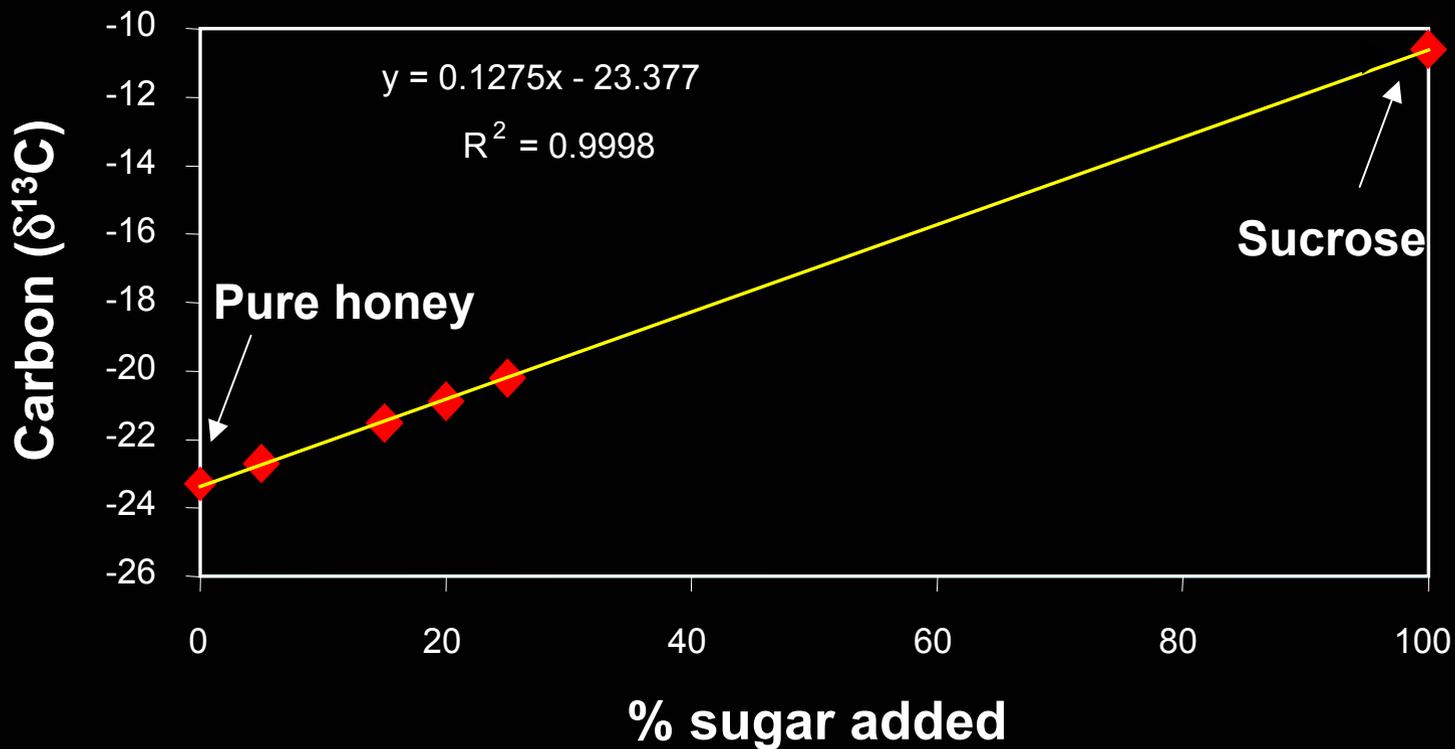


Adulterated honey

- ◆ Natural / Artificial
- ◆ Species
- ◆ Deliberate Adulteration
- ◆ Production Method
- ◆ Production Year
- ◆ Content
- ◆ Type, Variety
- ◆ Geographical origin
- ◆ Organic / Conventional

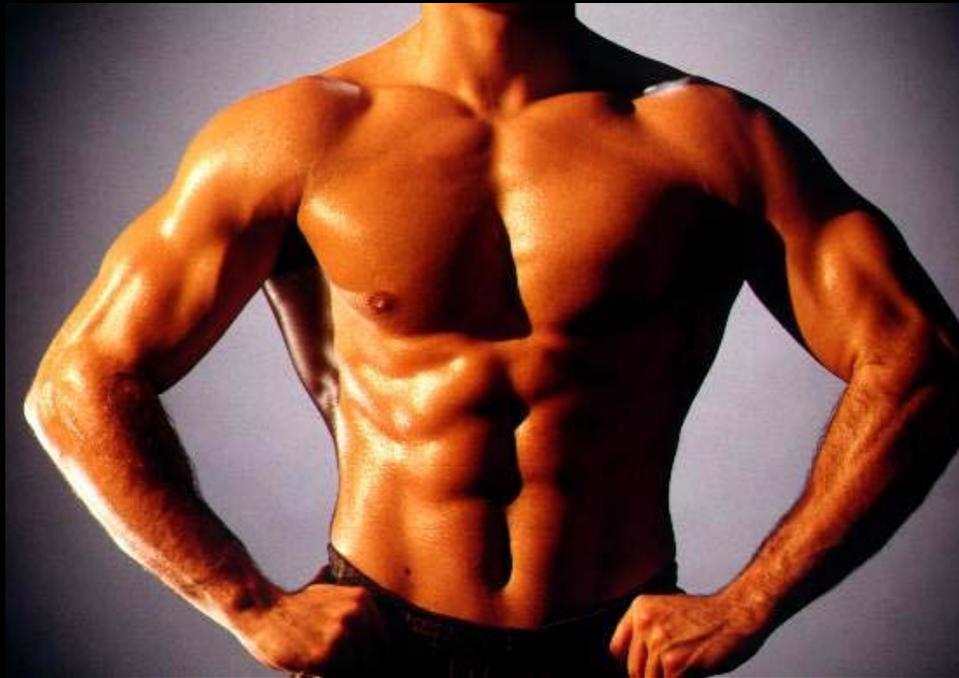


# Detection of adulterated honey using EA-IRMS



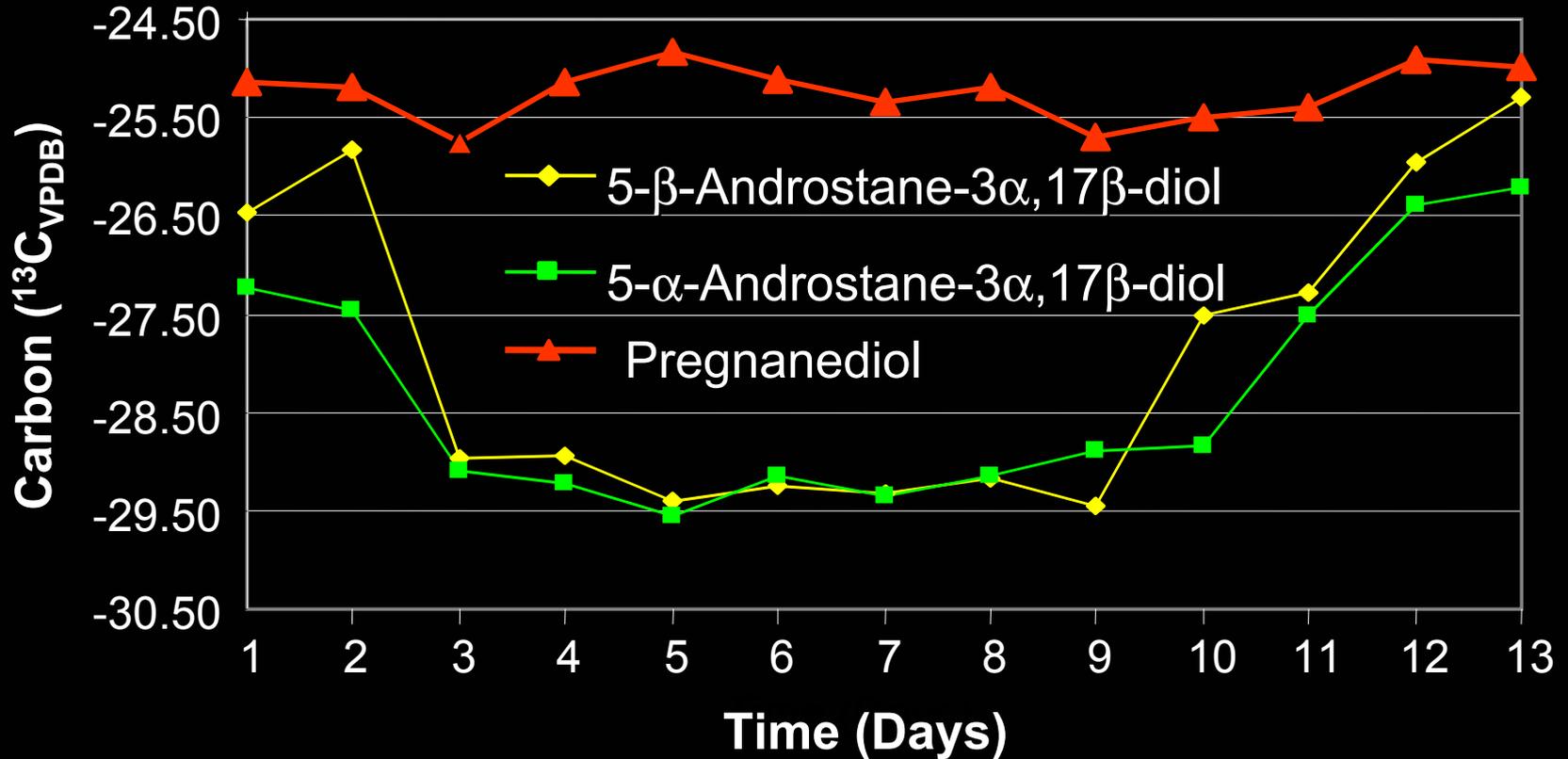


# Natural/synthetic: Doping in Sport





# GC-IRMS for Doping Control

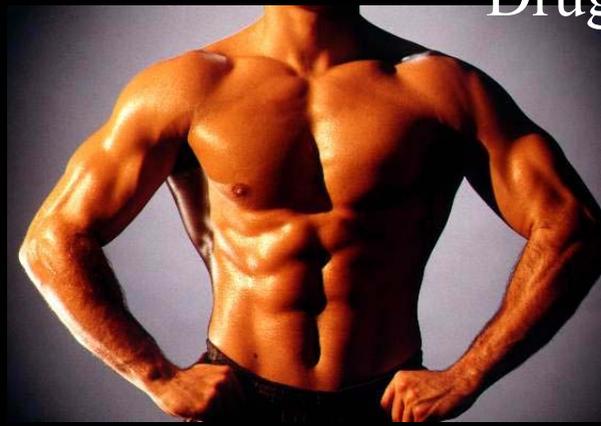


Testosterone administered at the end of day 2



# Misuse = proven unambiguously using IRMS

## Drug misuse: Testosterone



+ administered testosterone

	T/E ratio	IRMS
Volunteer 1	Negative	Positive
Volunteer 2	Positive	Positive
Volunteer 3	Positive	Positive
Volunteer 4	Negative	Positive
Volunteer 5	Negative	Positive

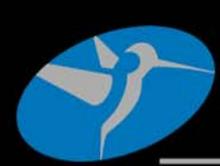
The affect of diet or time of administration can also be investigated by closely monitoring the ratio of the androstane diols to pregnane diols



# IRMS: IOC Recognised Method in Doping Control

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- ◆ GC-C-IRMS is the only technique which can unambiguously distinguish between endogenous and exogenous steroid metabolites
- ◆ Technique adopted by I.O.C.
- ◆ Micromass IsoPrime used for testing in International competitions: Winter Olympics in Nagano and Asian games in Thailand.



# Monarch Butterfly Migration

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# Butterfly Migration Patterns

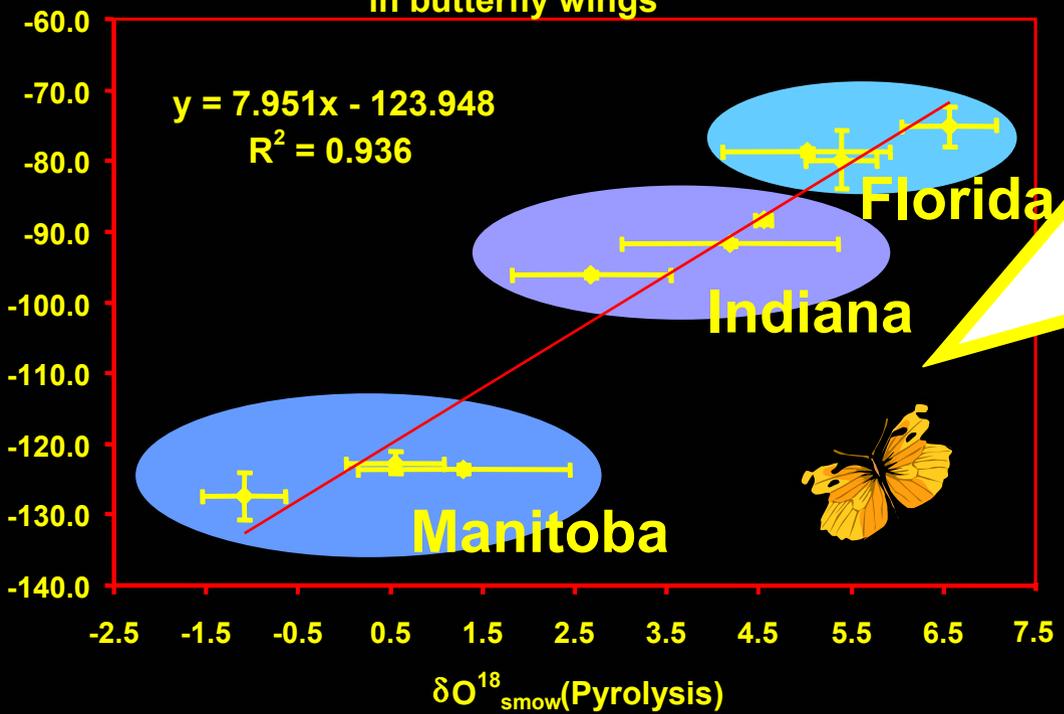


Map of North America showing sample sites and contours of weighted average  $\delta D$  (‰ SMOW) in summer rainfall (from IAEA Network 1958-1997)

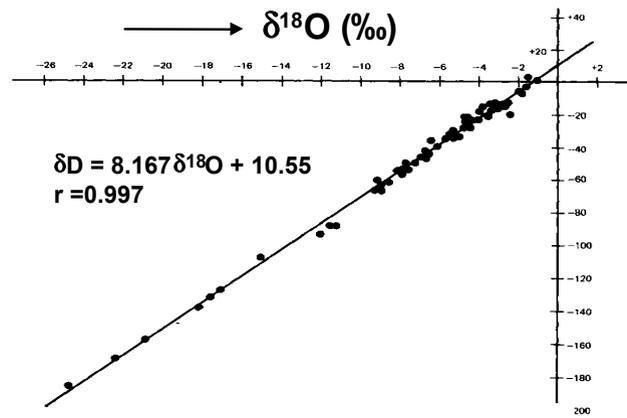


# Example of Application

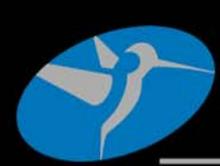
### Correlation between $\delta D$ pyrolysis and $\delta^{18}O$ pyrolysis in butterfly wings



### Atmospheric Waters

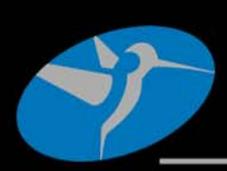


Relationship between weighted average  $\delta D$  and  $\delta^{18}O$  in precipitation  
 Ref. Stable Isotope Hydrology. Deuterium and Oxygen -18 in the Water Cycle.  
 IAEA, Vienna, 1981, pp. 121.



# Baleen Whale Migration Patterns EA-Pyrolysis





# Baleen Whale





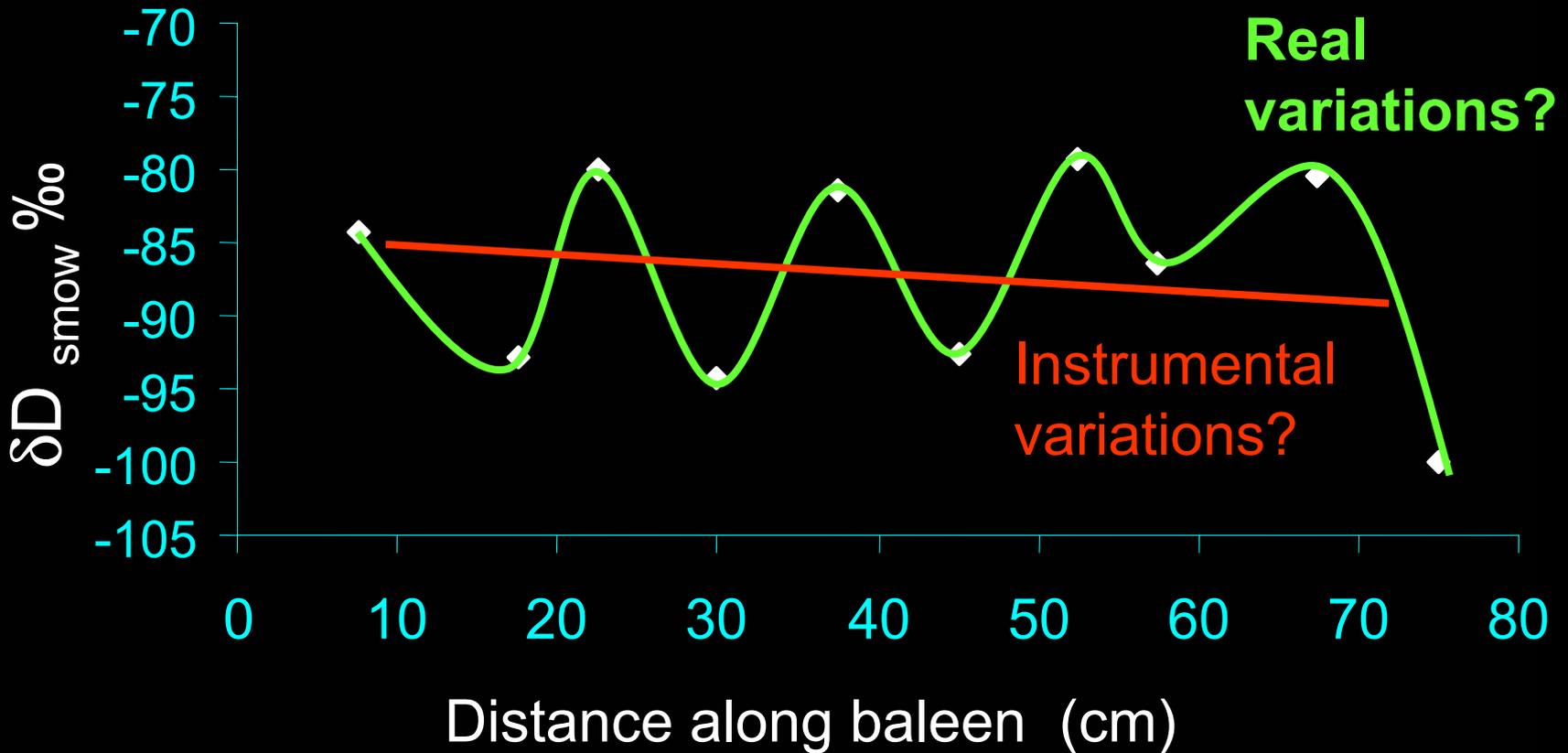
# Migration Route

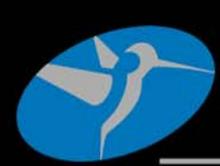




# Variation in $\delta D$ along the length of baleen

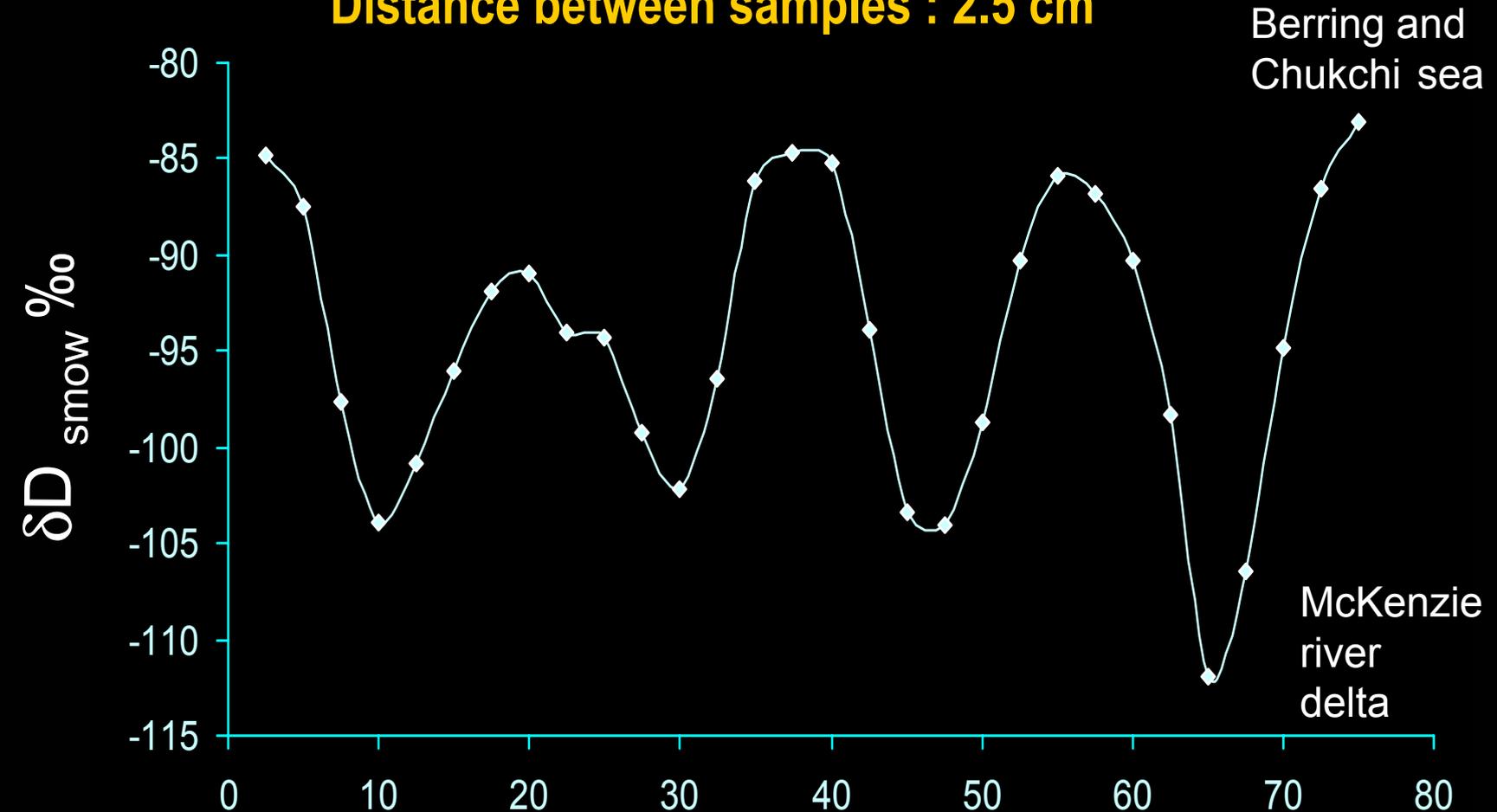
Distance between samples : 7.5 cm





# Variation in $\delta D$ along the length of baleen

Distance between samples : 2.5 cm



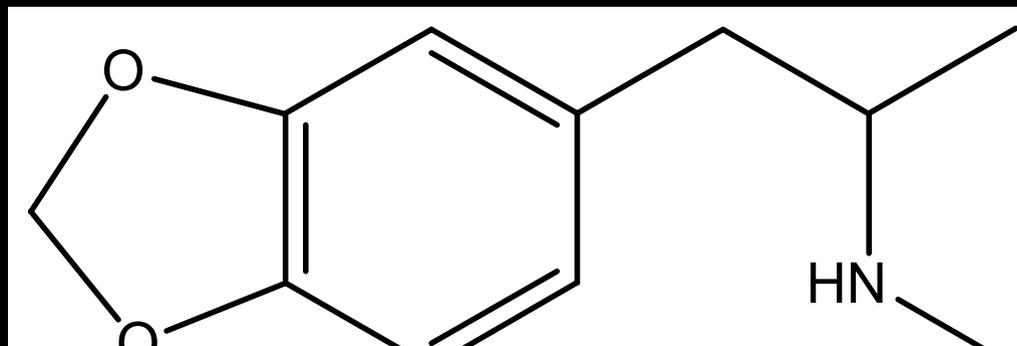


# IRMS and Ecstasy

F. Pahlol Univ. Nantes in Press



3,4-MethyleneDioxyMethAmphetamine





# Syntheses and precursors

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- Many possible synthetic pathways ...
- 6 commonly used by clandestine laboratories
- Main precursors:
  - 3,4-methylenedioxyphenyl-2-propanone (3,4-MDP-2-P)
  - safrole or isosafrole
- Main nitrogen sources
  - methylamine
  - formamide or N-methylformamide



# GC-IRMS $^{13}\text{C}$ analyses

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## - Precursors :

Sassafras : -28 to -27 ‰

Safrole : -27 to -25,5 ‰

Isosafrole : -28 to -25,5 ‰

3,4-MDP-2-P : -28 to -27 ‰

Piperonal : -28 to -26,5 ‰

## - Seizures :

Ecstasy : -29 to -23 ‰



# GC-IRMS $^{15}\text{N}$ analyses

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## - Precursors :

Methylamine HCl : 2 groups: -5 and -0,5 ‰

Methylamine aq : 2,8 to 5,5 ‰

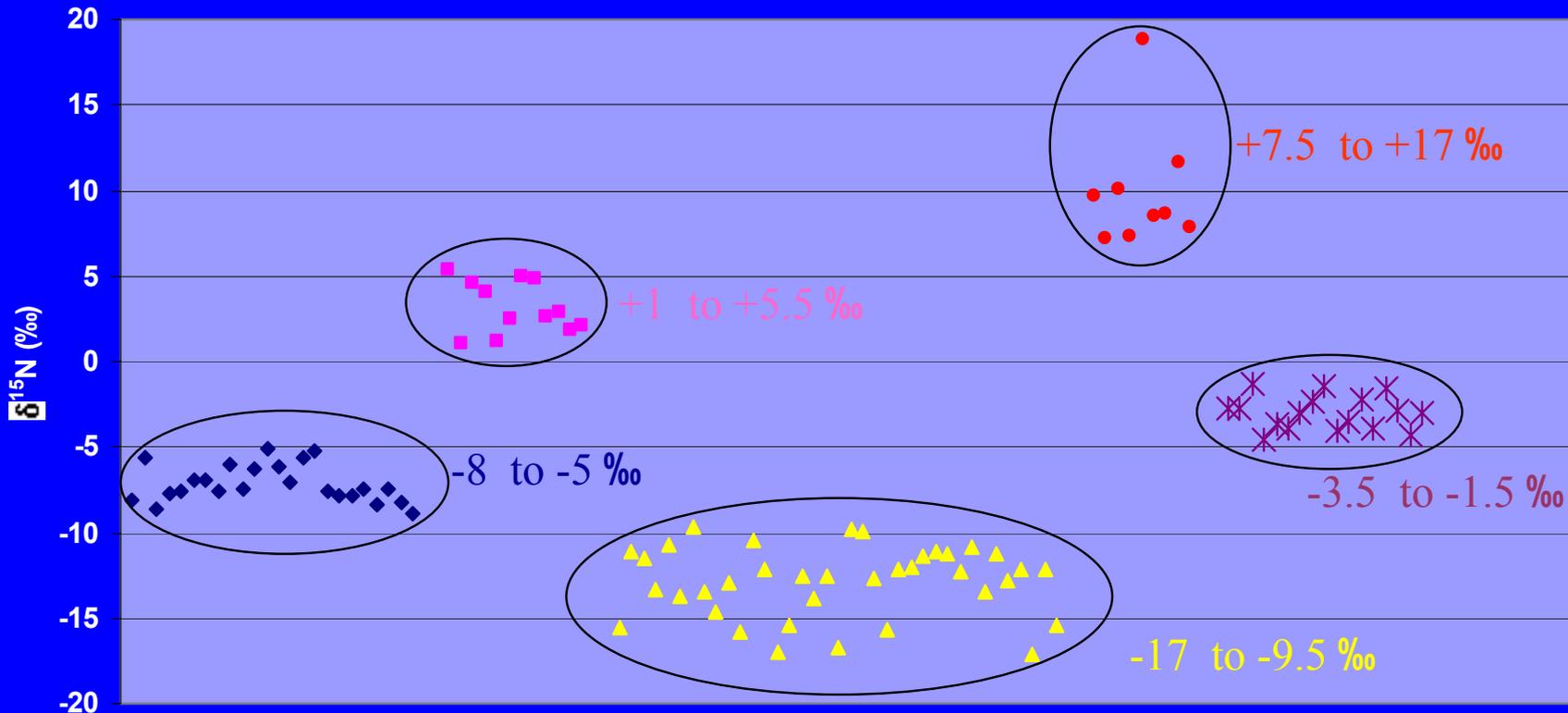
Formamide : -1,5 to -0,5 ‰

N-methylformamide : 2 groups: -5 and -1 ‰

## - Seizures :

Ecstasy : between -17 and 19 ‰

# Results for MDMA tablets ( $\delta^{15}\text{N}$ )





# Ecstasy seizures



2 seizures in North of France, Dec. 01 and Jan. 02

$\delta^{15}\text{N} = -7,56$  and  $-7,89$  ‰

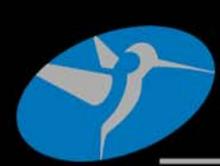
8 seizures (27,000 tablets) in 2001...

5 seizures had  $\delta^{15}\text{N}$  between  $-12,25$  and  $-12,68$  ‰



2 seizures (30,000 tablets): Paris and Spanish frontier

$\delta^{15}\text{N} = -8,61$  and  $-8,67$  ‰



## IRMS and heroin



From Desage et al 1991  
F. Besacier PhD thesis 1999



# Drug classification

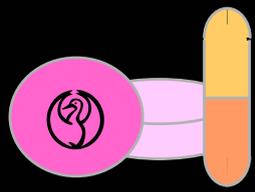


➔ Natural origin



- Plants (Cannabis)
- Substances extracted from plants (Cocaine)
- Partially synthetic drugs (Heroine)

➔ Synthetic origin



- Amphetamine and by-products



# Heroin manufacturing



*Papaver Somniferum L. var. Album*



Opium



Morphine



Diacétylmorphine



# Why IRMS?

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## Geographical origin

- Natural drugs

## Synthetic pathways

- Synthetic drugs
- Partially synthetic drugs

## Manufacturing batch

- All drugs

## Mixing

- All drugs except natural



# Application to heroin

## *Pavot*

Morphine  
Codéine

$^{13}\text{C}/^{12}\text{C}$   
 $^{15}\text{N}/^{14}\text{N}$



geographical origin



Acétylation

Diacétylmorphine  
Acétylcodéine

$^{13}\text{C}/^{12}\text{C}$   
 $^{15}\text{N}/^{14}\text{N}$



geographical origin

+

synthesis

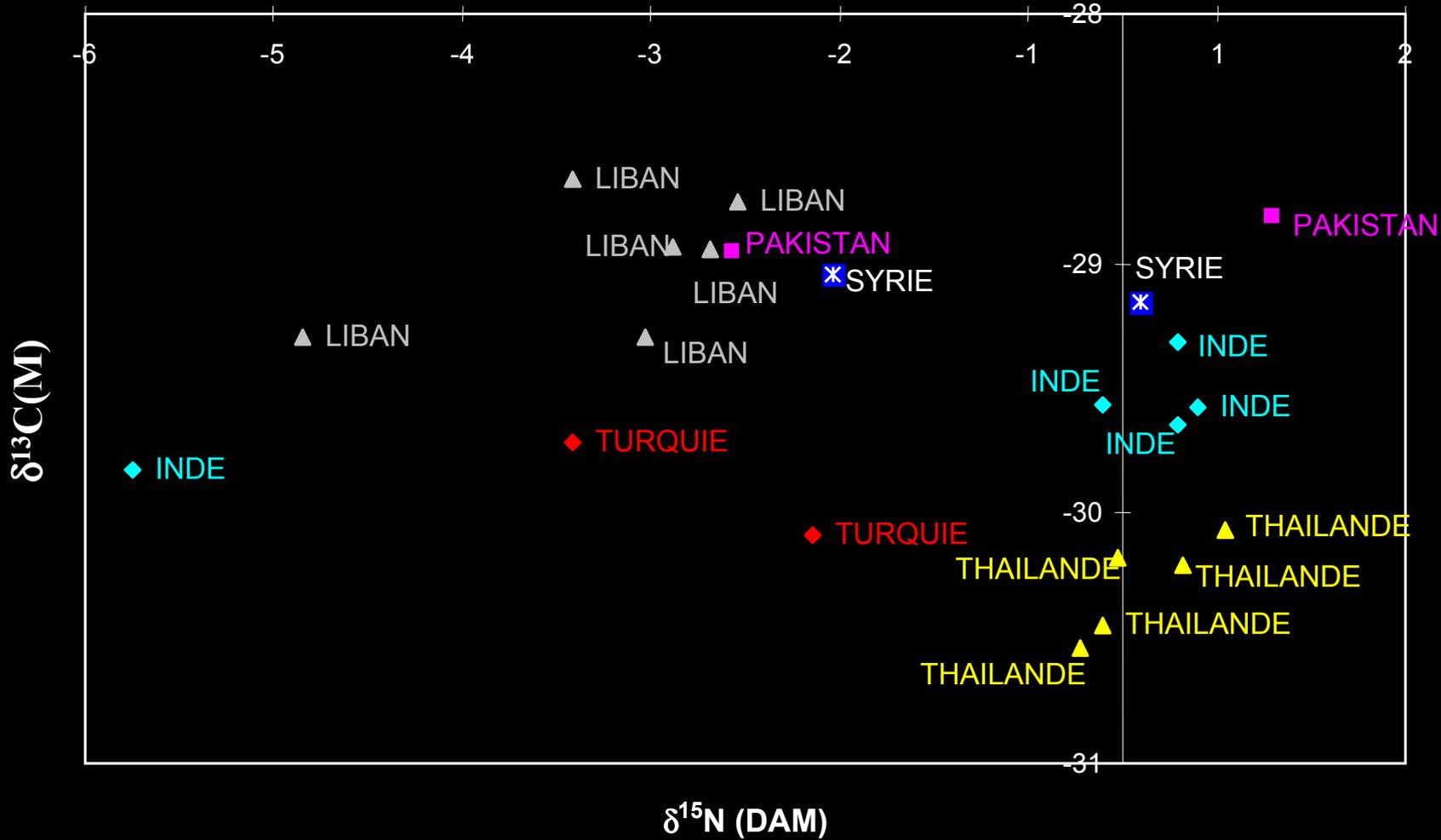


geographical origin

## *Héroïne*



# Geographical origin





# Synthetic pathway

Morphine  
Codéine

$^{13}\text{C}/^{12}\text{C}$



geographical origin



Hydrolysis



différence  
||  
synthesis

Diacétylmorphine  
Acétylcodéine

$^{13}\text{C}/^{12}\text{C}$

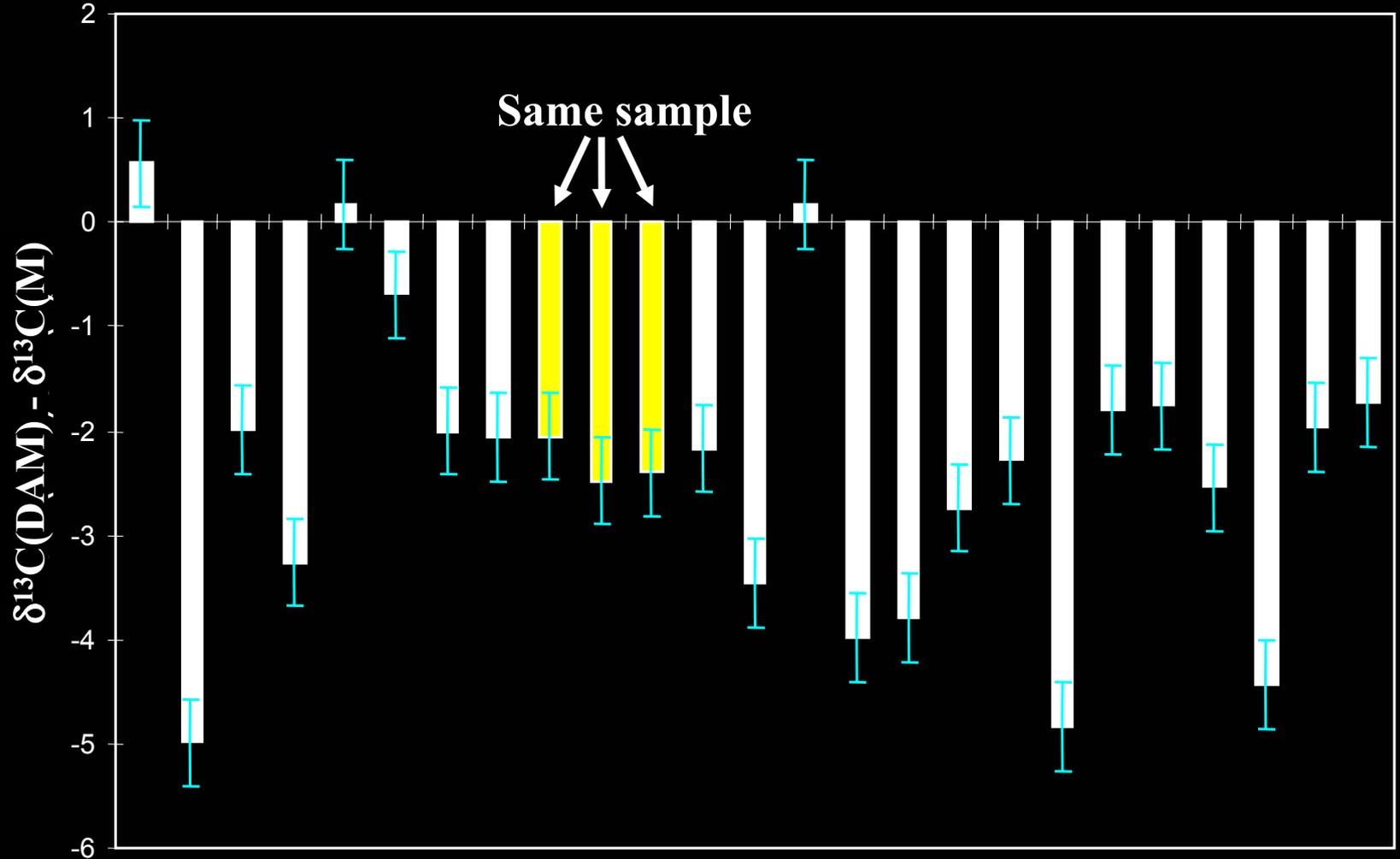


geographical origin  
+  
synthesis

*Héroïne*



# Synthetic pathway





# IRMS and Drug Product Authentication

Drug Batch Characterisation Realised Through  
New Generation High Throughput IRMS



Collaboration with J. Jasper MIT.



# The Cost of Counterfeiting

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**"Counterfeit paracetamol [acetaminophen] syrup cost the lives of 109 children in Nigeria, and whilst Third World countries are undoubtedly the most vulnerable to attack, there is clear evidence of highly organized criminal activity affecting most of the developed world."**

From Geoff Power, 1999 ICPR No. 476 - 477



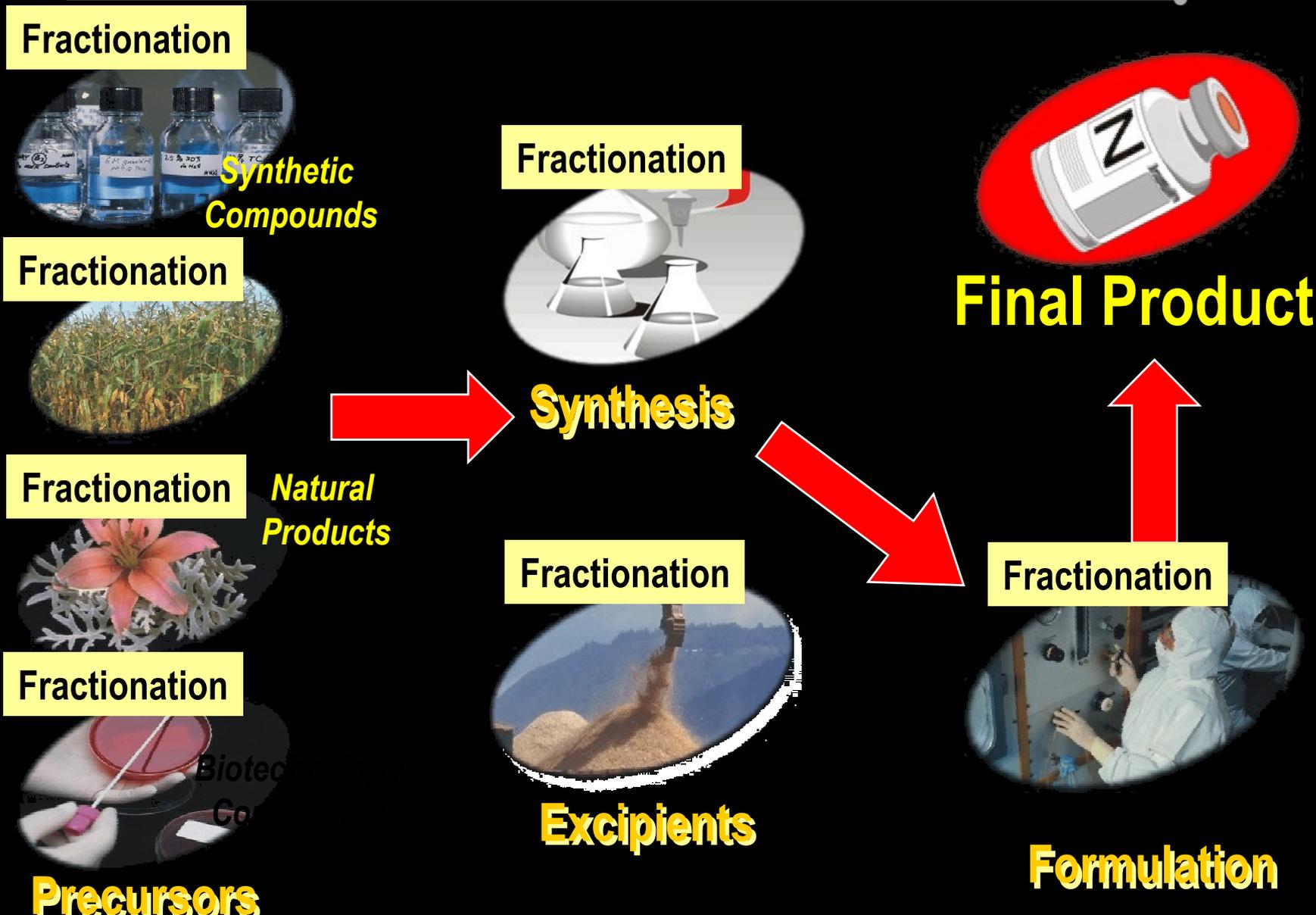
# Why use Stable Isotopes?

- ◆ Using natural abundance isotopic ratios, no spiking is required
- ◆ Unique signature, expected to be highly variable between batches of the same drug - thus forming the basis for the technique
- ◆ Using GC and EA IRMS we have an effective way of targeting active compounds and measuring, very precisely, their isotopic ratios
- ◆ Highly effective multidimensional isotopic approach: H, C, N, O, S





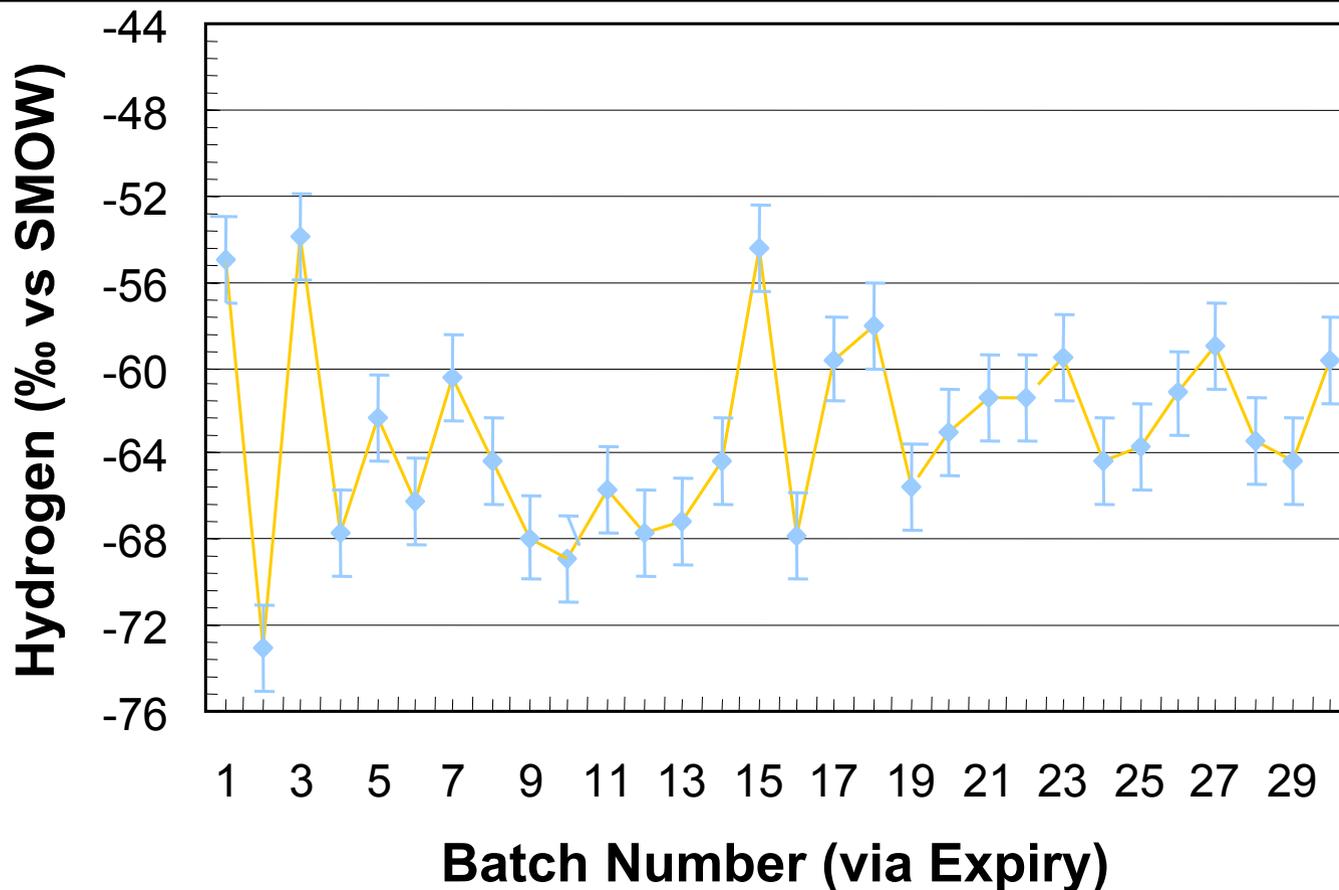
# Factors involved in Inter-Batch Variability



# Clear Temporal Variation in Analgesic Products

Hydrogen

Acetaminophen 1





# What can we detect?

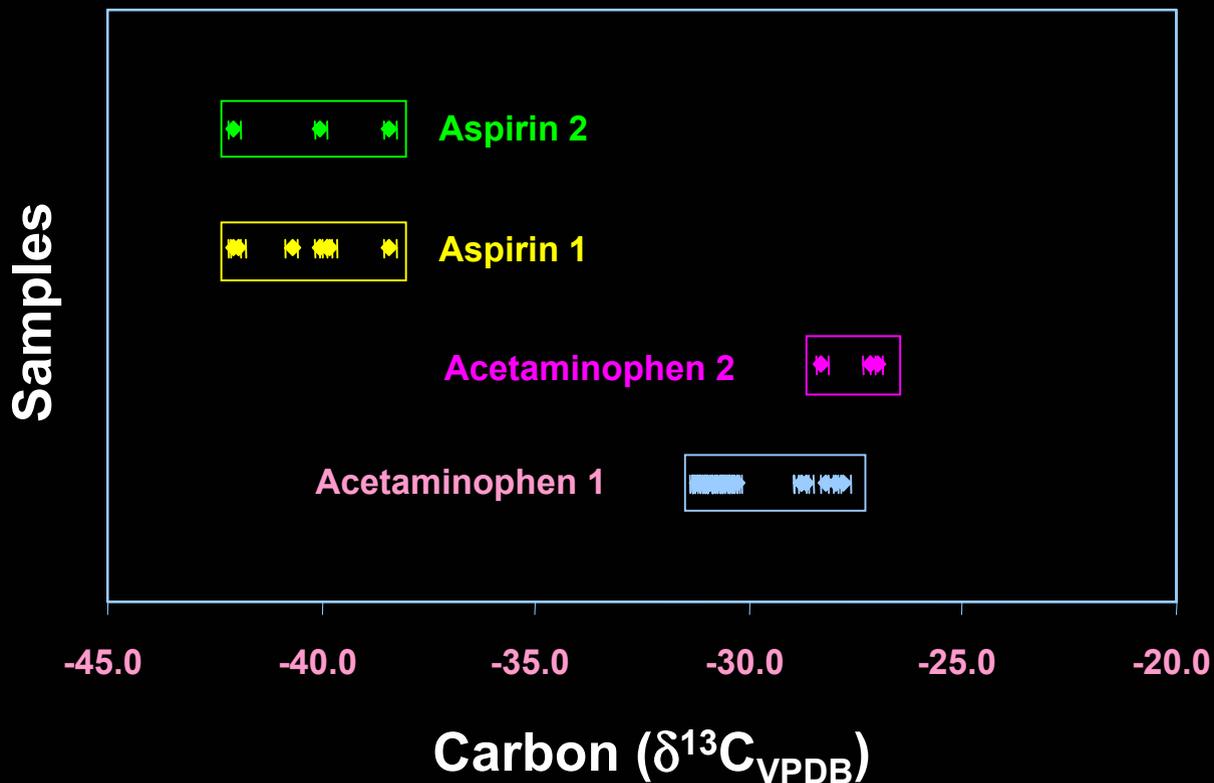
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- ◆ Ext. Reproducibility  $\Rightarrow$   $^{13}\text{C}$  0.2 ‰
- ◆ Ext. Reproducibility  $\Rightarrow$   $^{18}\text{O}$  0.3 ‰
- ◆ Ext. Reproducibility  $\Rightarrow$  D/H 2 ‰
  
- ◆ Within batch variations in the order of Ext. reproducibility of the methods



# Significant Inter-Batch Isotopic Variability

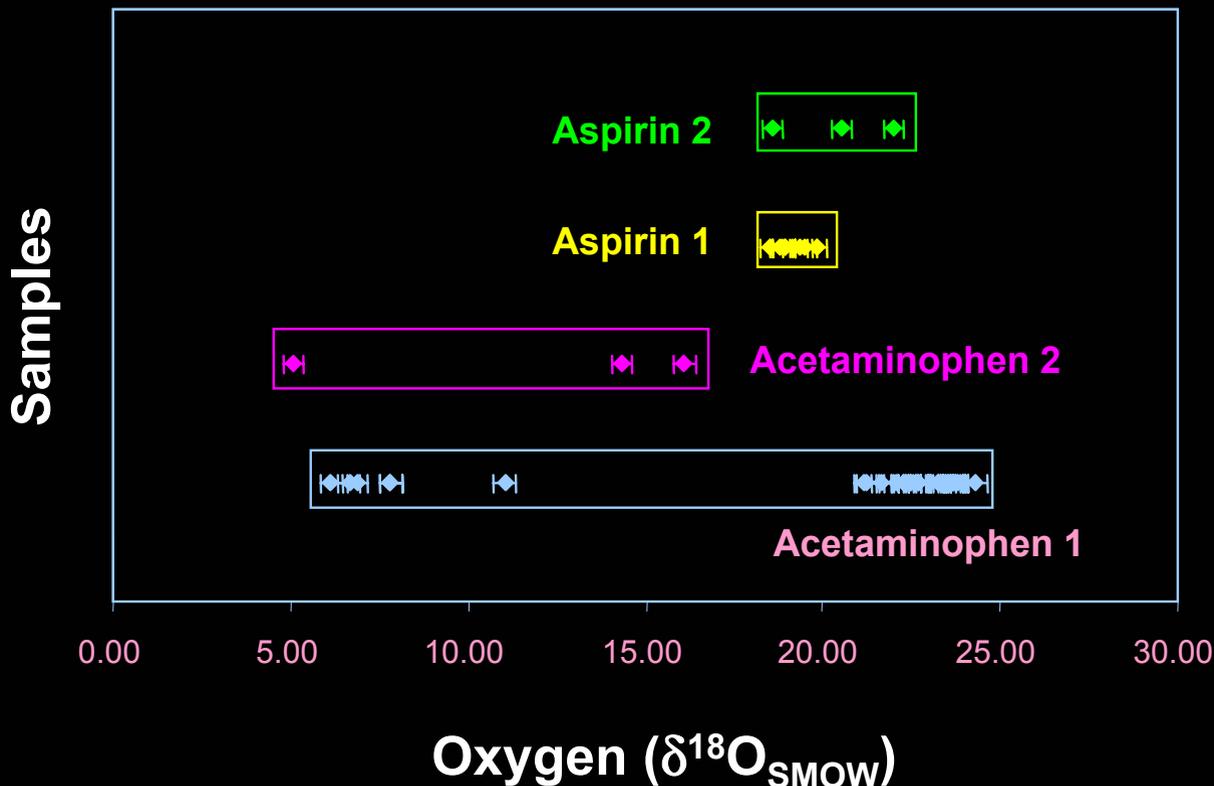
## Carbon





# Significant Inter-Batch Isotopic Variability

## Oxygen



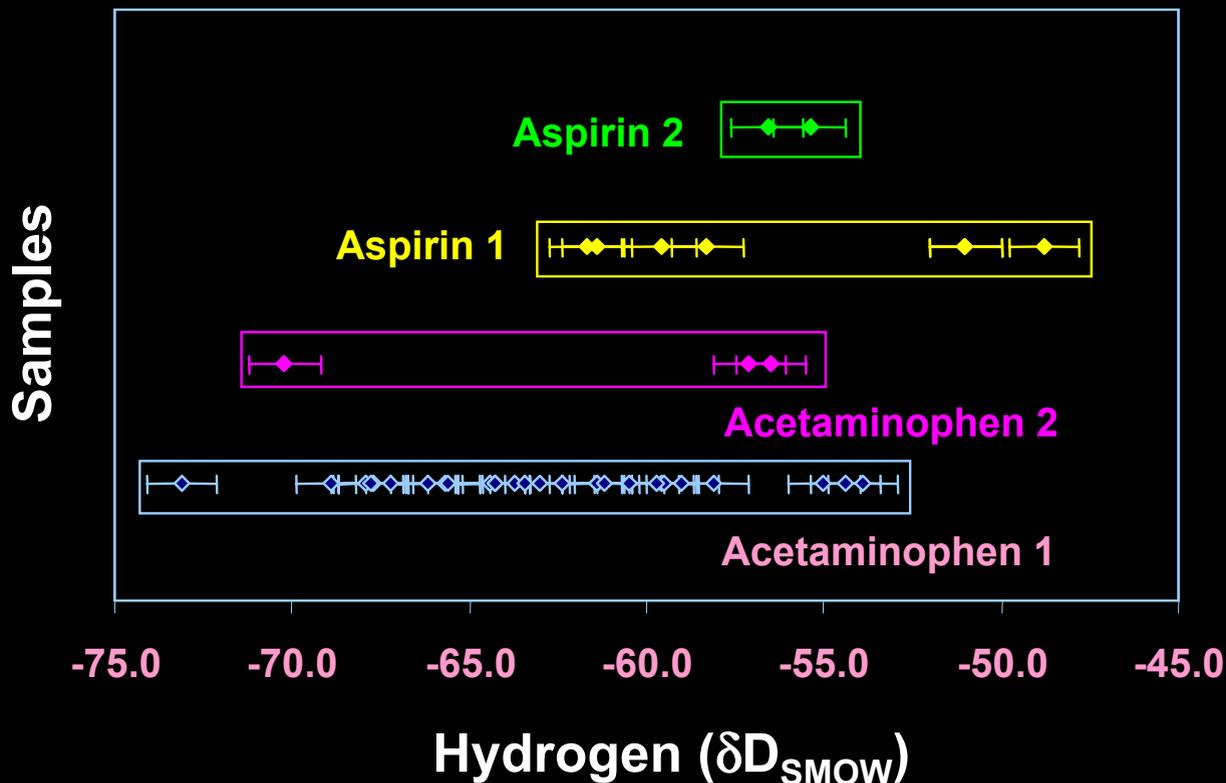
Analyses of 3 different samples from the same batch

$\delta^{18}\text{O}_{\text{SMOW}} = 7.91$ , StDev = 0.21



# Significant Inter-Batch Isotopic Variability

## Hydrogen

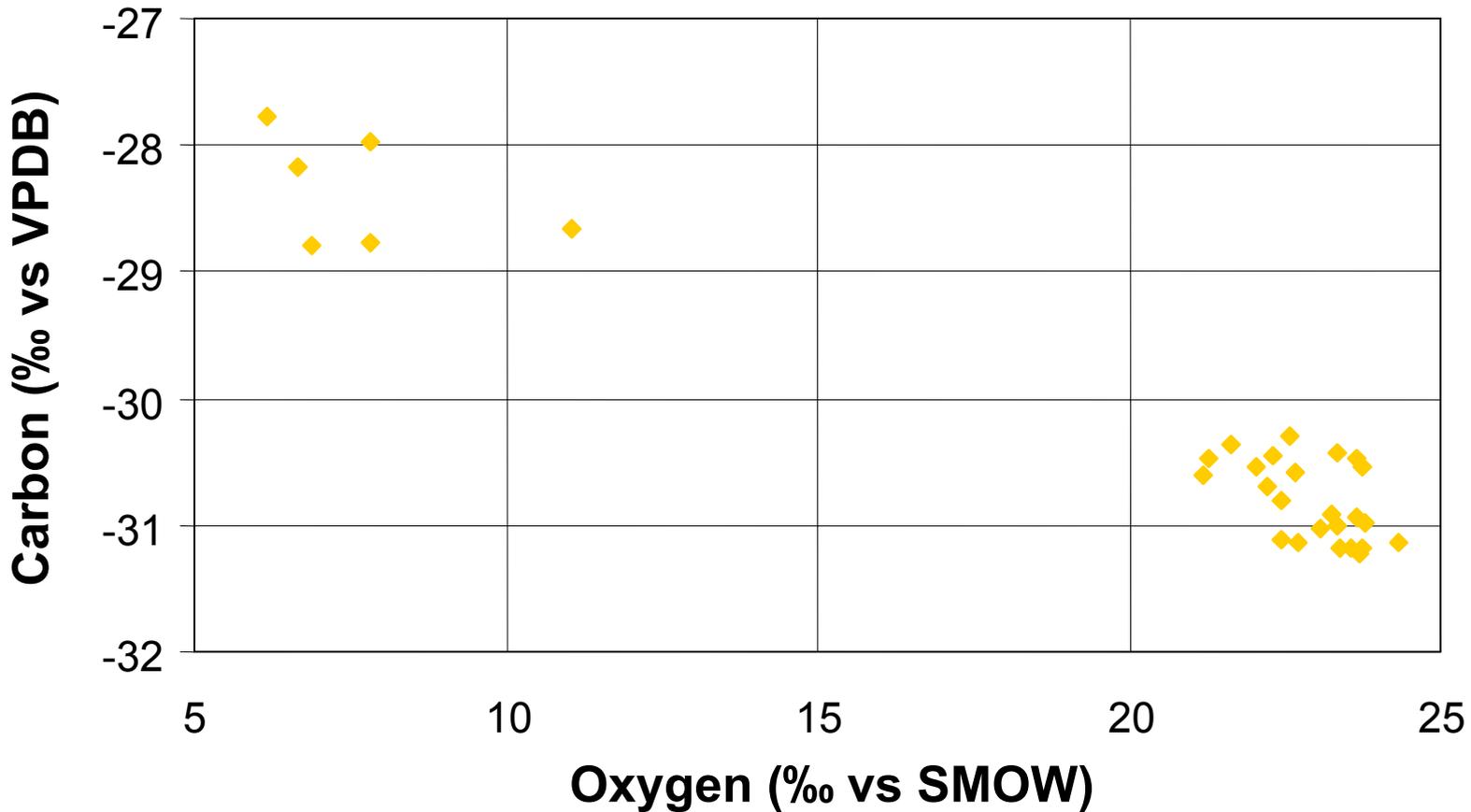




# One Product, Different Batches, Large Variation

## Oxygen vs Carbon

Acetaminophen 1





# Product Authentication

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- ◆ Characterisation of the ambient batch to batch stable isotopic variation of such products represents a means by which to 'fingerprint' individual drug batches
- ◆ By bar-coding products with their specific isotopic 'fingerprint' we make them very difficult/impossible to counterfeit
- ◆ Active compound authentication ultimately aids control of the supply, use and disposal of drugs, saving money, reputation and lives



# CONCLUSION

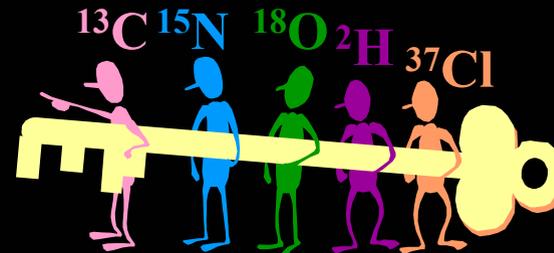
## Isotopic ratios (EA-IRMS, GC-IRMS)



Precious tool for police and justice

- geographical origin
- synthetic pathway
- Natural/synthetic origin
- batch identification

## Origin determination :



## Used in combination with other methods

