

# Isotopic Characterisation of Amphetamines

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1- School of Chemistry, University of Bristol

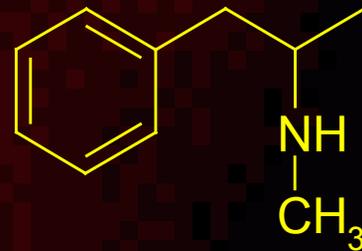
2- Mass Spec Analytical Ltd, Bristol



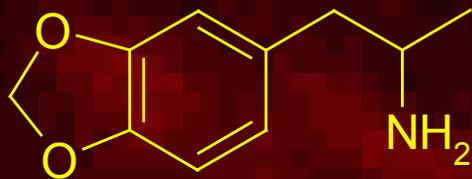
# Amphetamines



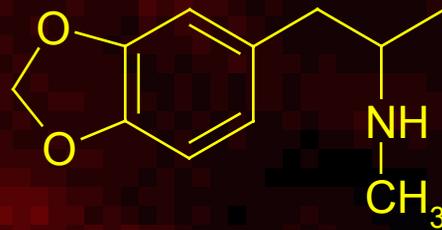
amphetamine



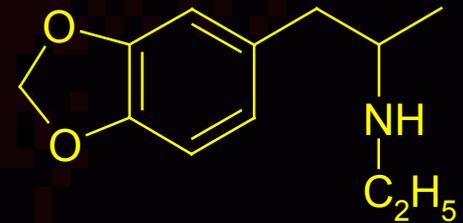
methamphetamine



MDA

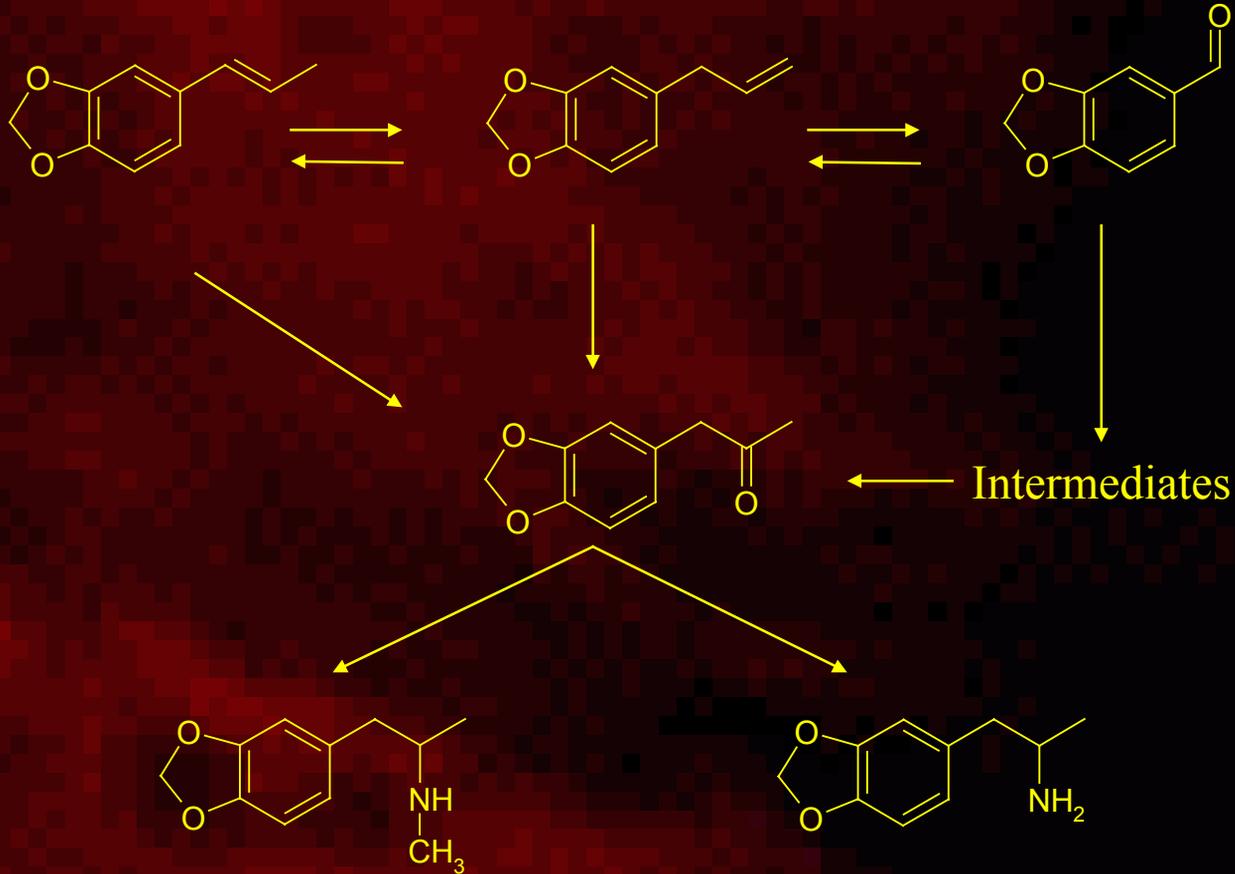


MDMA



MDEA

# Synthesis



# Overall Aim

- Use “isotopic fingerprint” or “DNA” to:
- Distinguish **Batch – Batch**
- Link “**trace**” and “**bulk**” evidence
- Trace drugs to a common source of manufacture or supply
- Determine synthetic method

# Why Bother ?

- Home Office statistics for 1999:
- 29% of 16-29 year olds have experienced hallucinogenic drugs
- Police and HM Customs seized approximately 6.5 million ecstasy tablets
- Estimated supply of 26 million tablets

# Study I

Five batches of “ecstasy” tablet  
supplied by Avon & Somerset Constabulary, Scientific Investigations



RN/2932/99



RN/1491/00



RN/6108/00

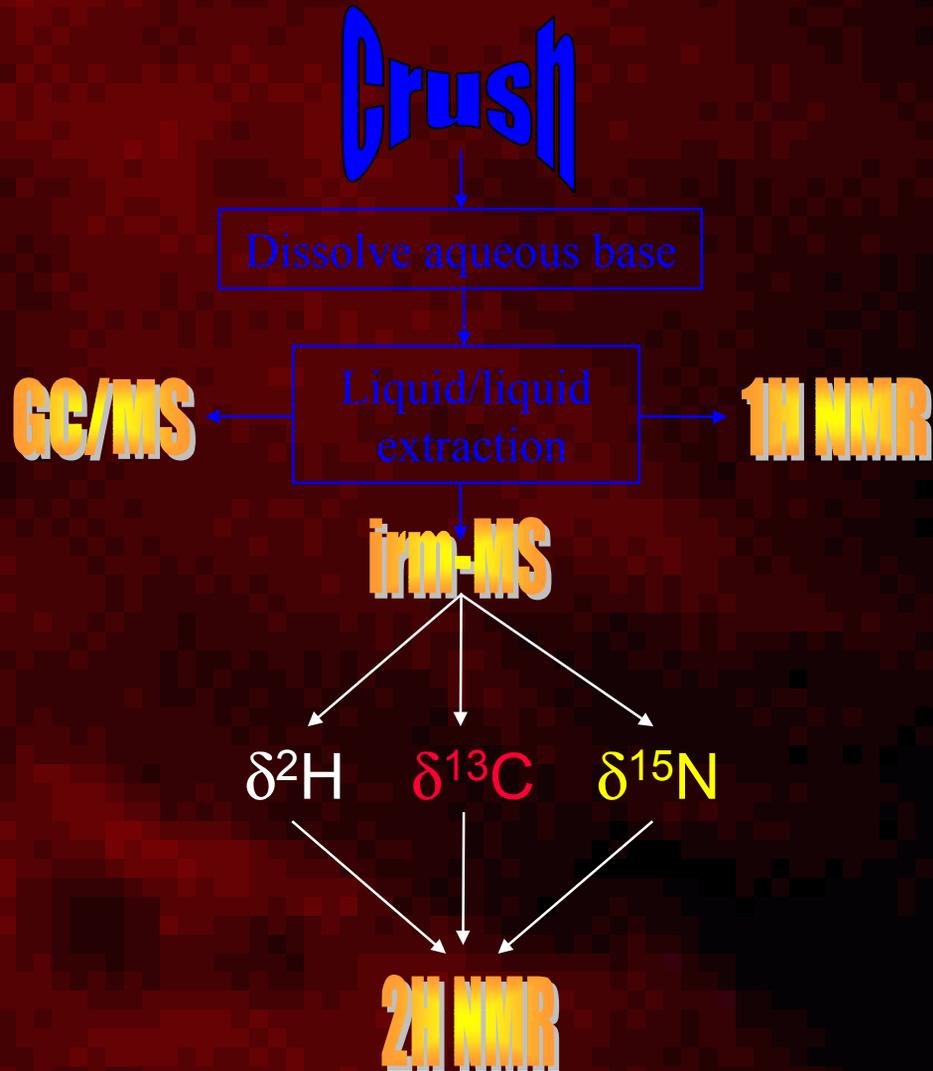


RN/883/00



RN/1061/00

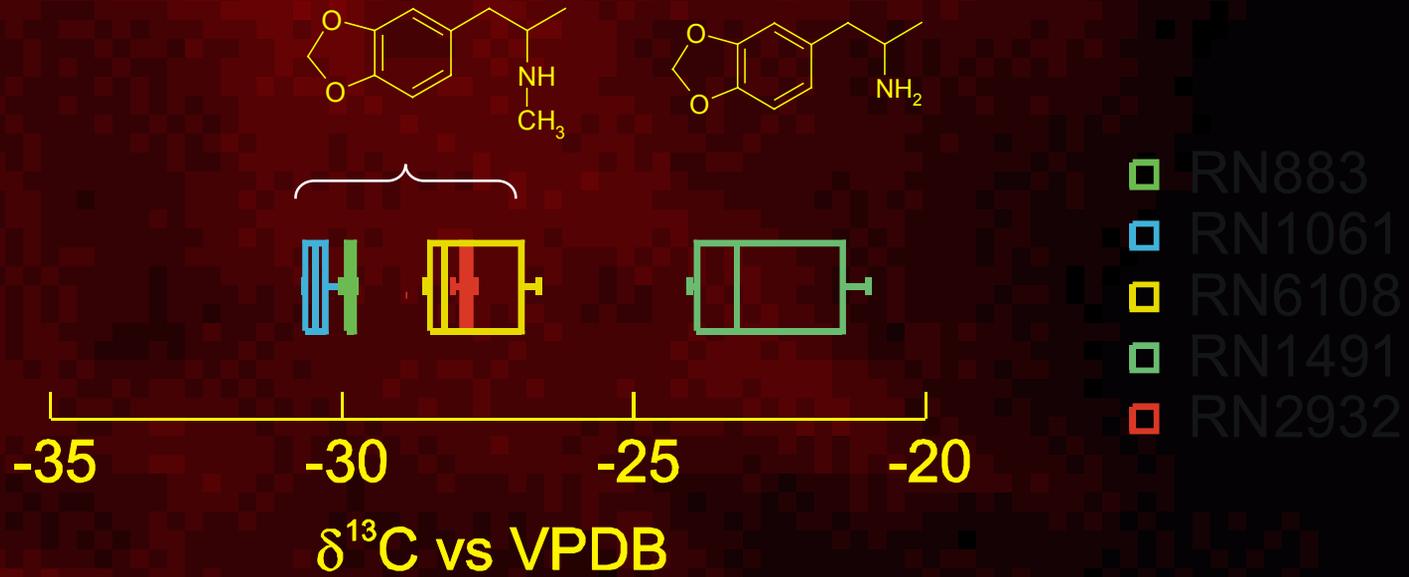
# Methodology



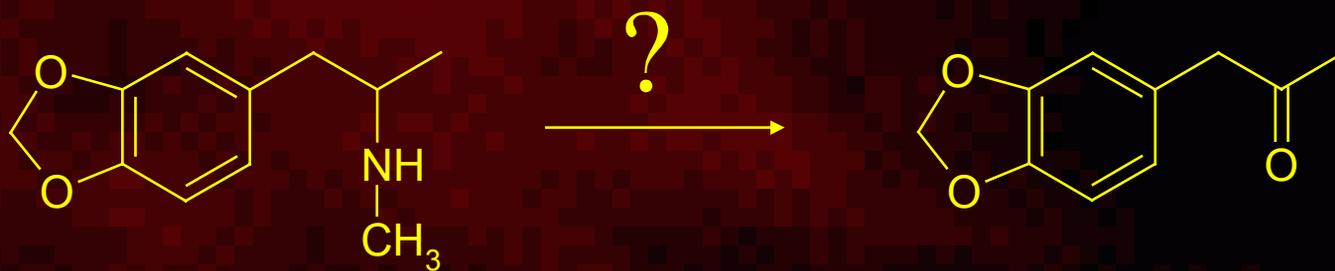
# $\delta^{13}\text{C}$ analysis

- ThermoFinnigan Delta<sup>PLUS</sup>XL
- GCC III interface
- Cu/Ni/Pt oxidation reactor 950 °C
- Cu reduction reactor 600 °C
- Calibrated to NIST sucrose (-10.47‰ vs VPDB)

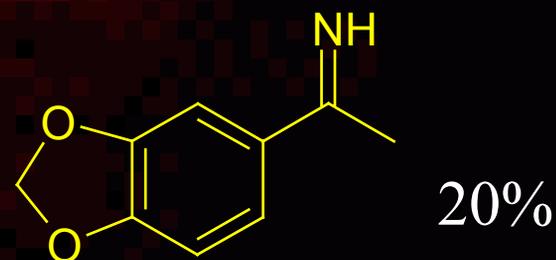
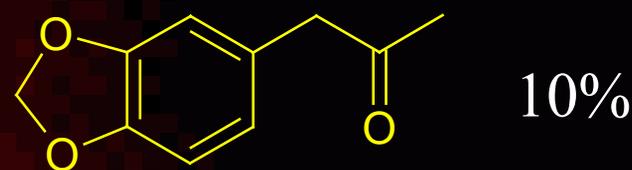
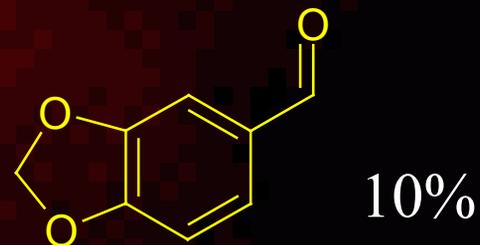
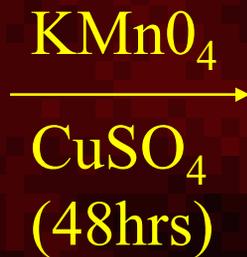
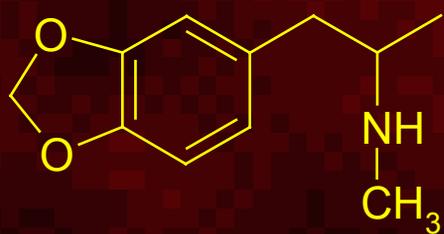
# $\delta^{13}\text{C}$ results



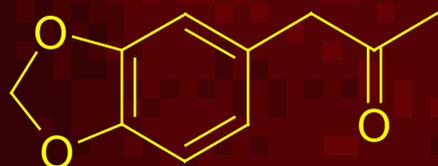
# Aside



# Aside



# Aside

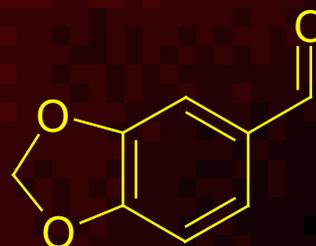


-26.95‰±0.24

-26.51‰±0.27

-26.85‰±1.05

-25.03‰±0.23



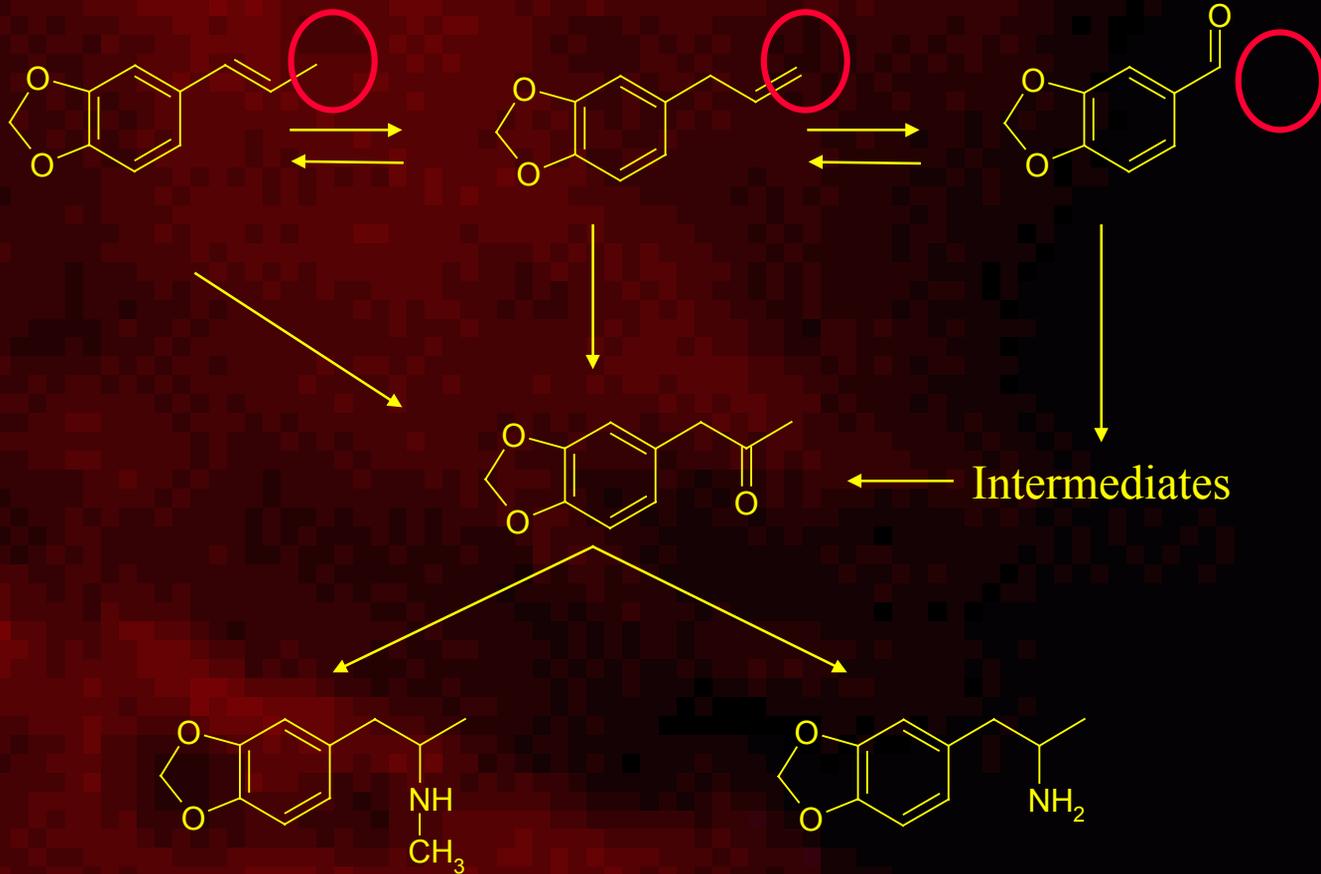
-27.28‰±0.08

-26.14‰±0.32

-26.56‰±1.15

-22.11‰±0.25

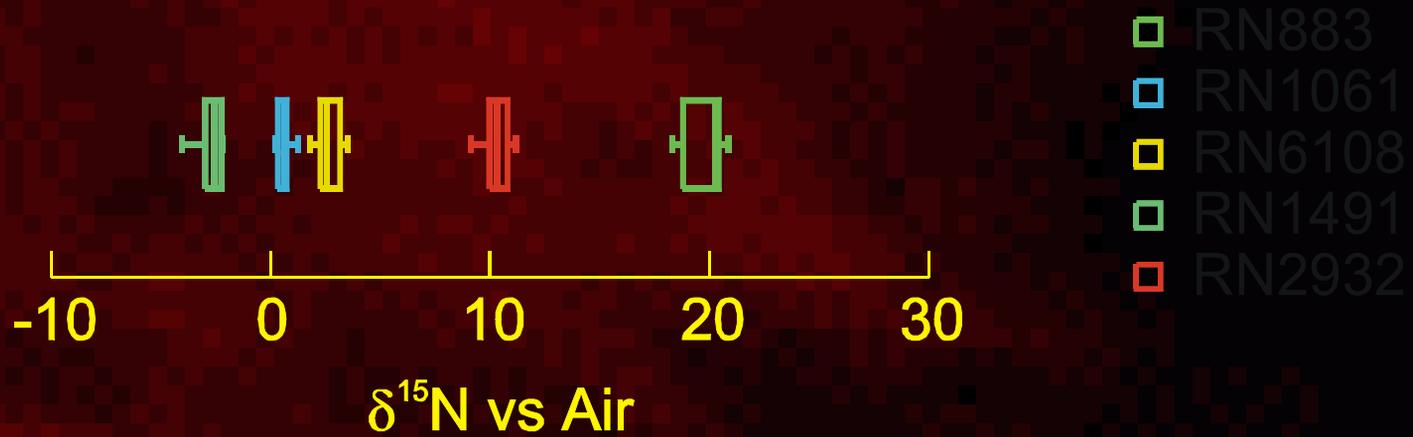
# Synthesis



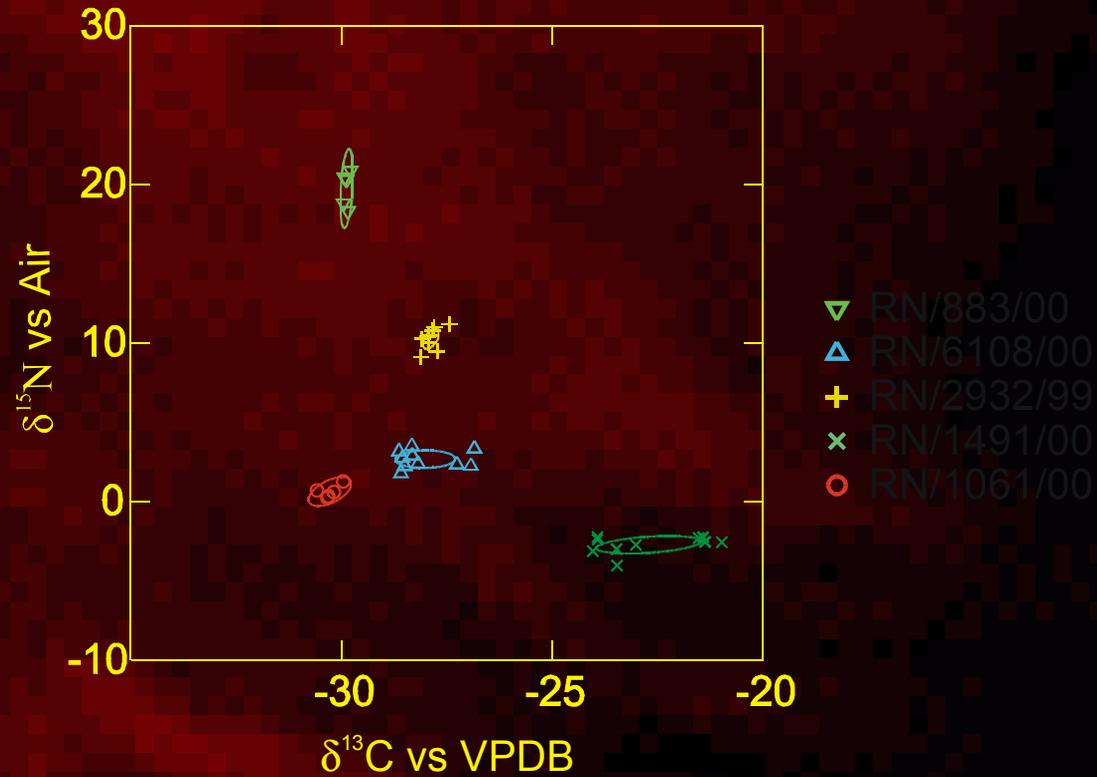
# $\delta^{15}\text{N}$ analysis

- CarloErba 2500 EA  
ThermoFinnigan Delta<sup>PLUS</sup>XL
- Cr oxide and Ag/Co oxide at 1050 °C
- Cu reduction at 640 °C
- Calibrated to NIST ammonium sulphate  
(0.4‰ and 20.3 ‰ vs air)

# $\delta^{15}\text{N}$ results



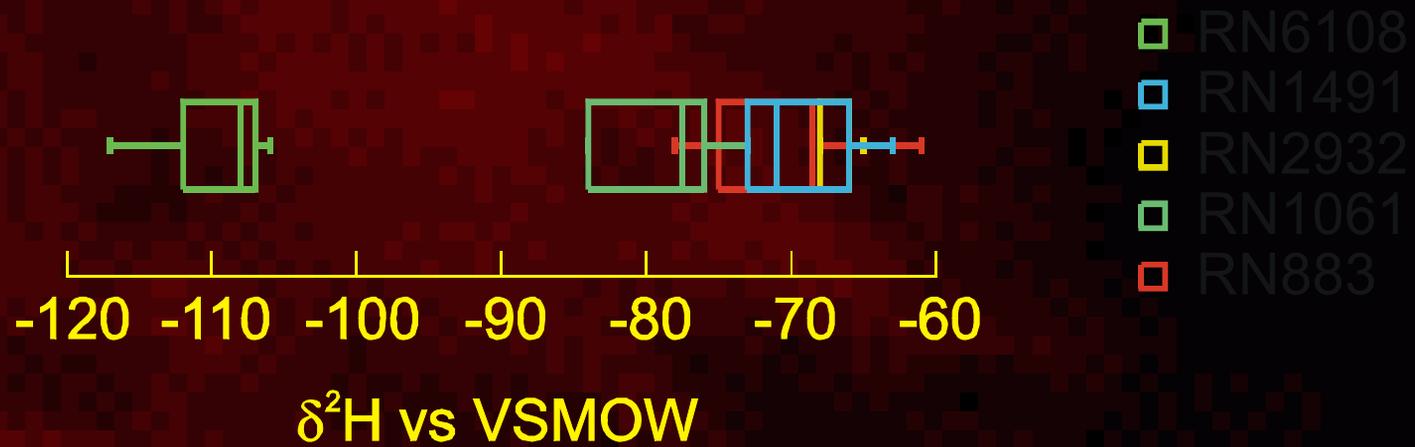
# $\delta^{13}\text{C}/\delta^{15}\text{N}$ results



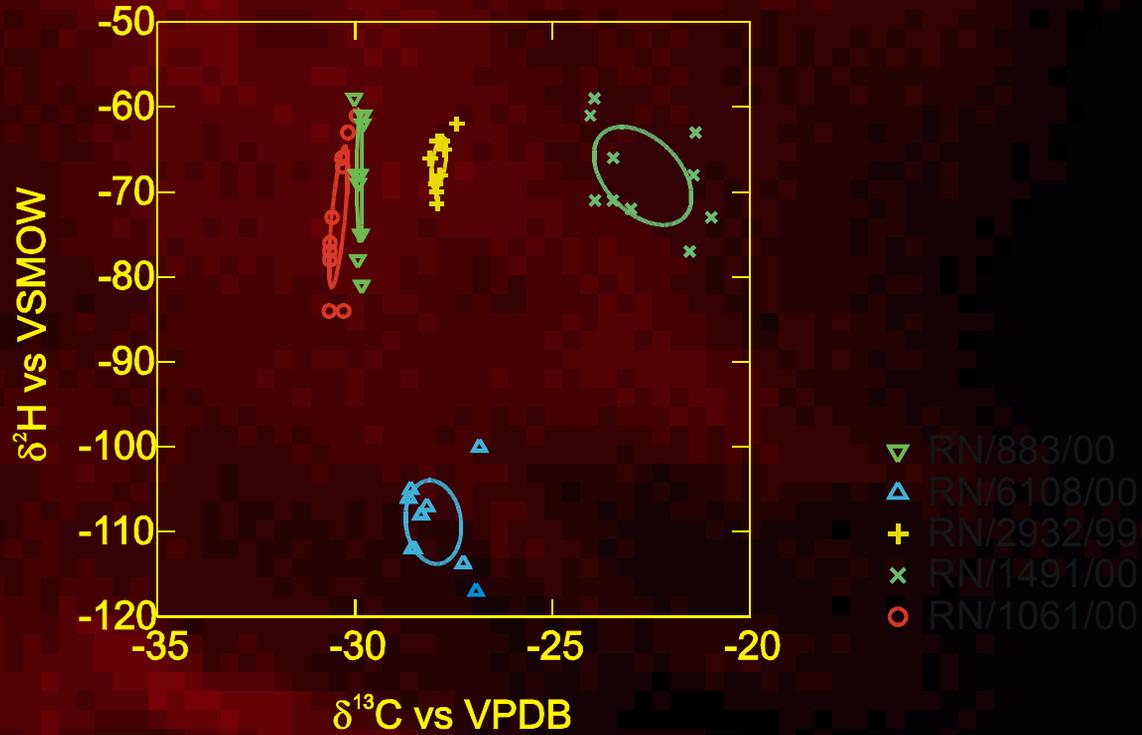
# $\delta^2\text{H}$ analysis (n=9)

- ThermoFinnigan Delta<sup>PLUS</sup>XL GC-TC-IRMS
- Thermal conversion at 1450 °C
- Calibrated to a suite of n-alkanes  
(C<sub>16</sub> - C<sub>30</sub> - calibrated against VSMOW)

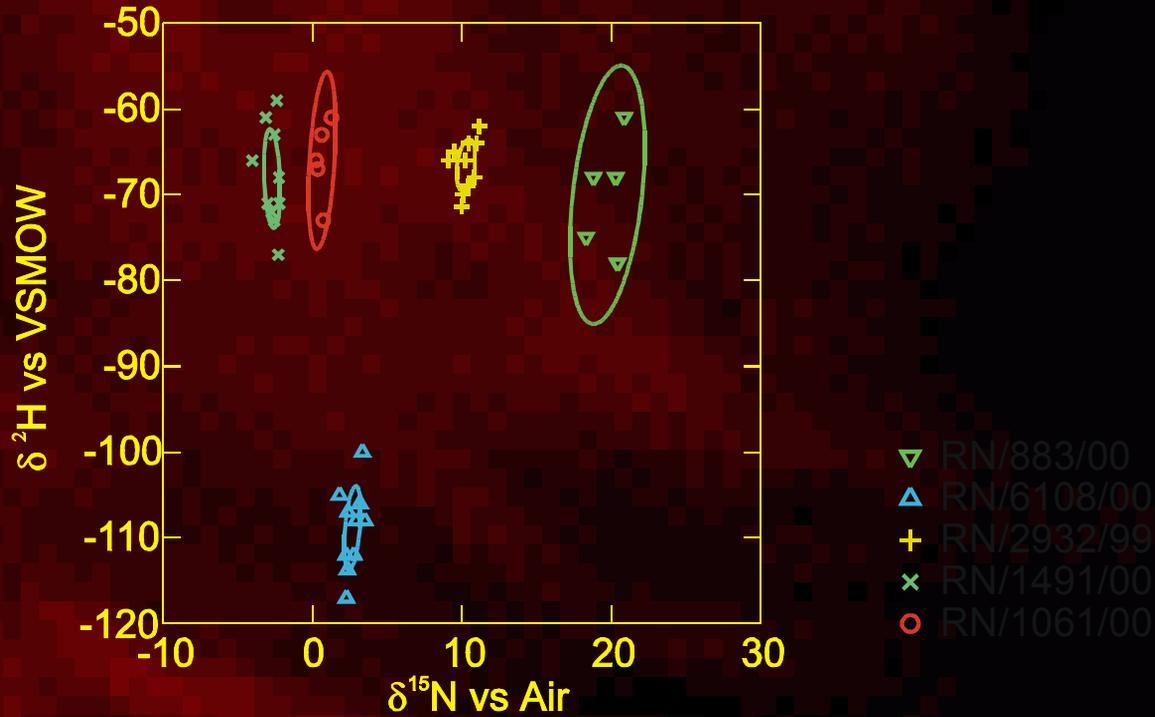
# $\delta^2\text{H}$ results



# $\delta^2\text{H}/\delta^{13}\text{C}$ results

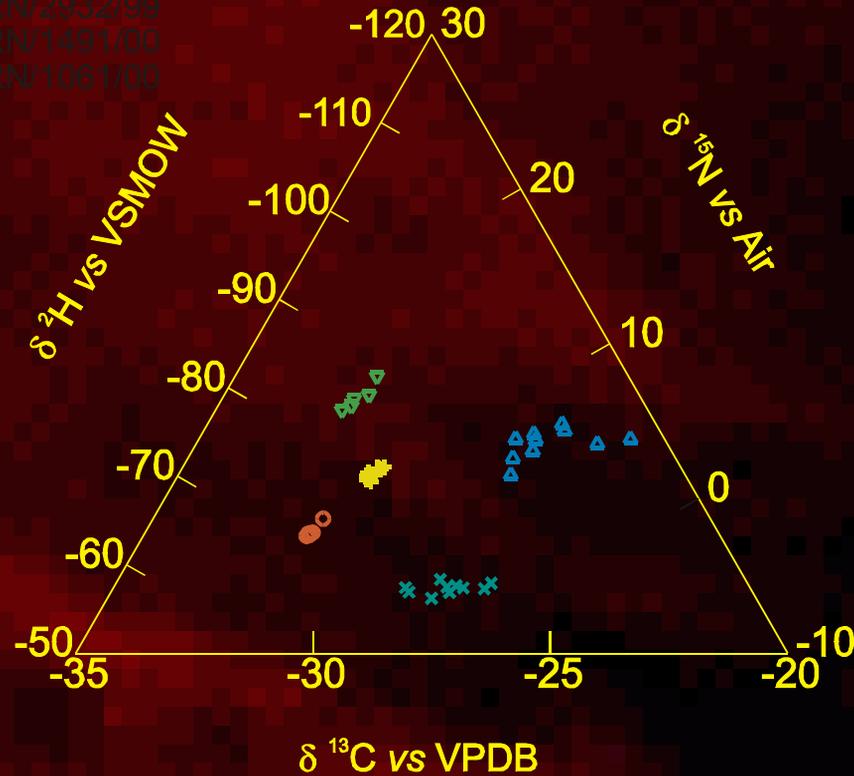


# $\delta^2\text{H}/\delta^{15}\text{N}$ results



# $\delta^2\text{H}/\delta^{13}\text{C}/\delta^{15}\text{N}$ results

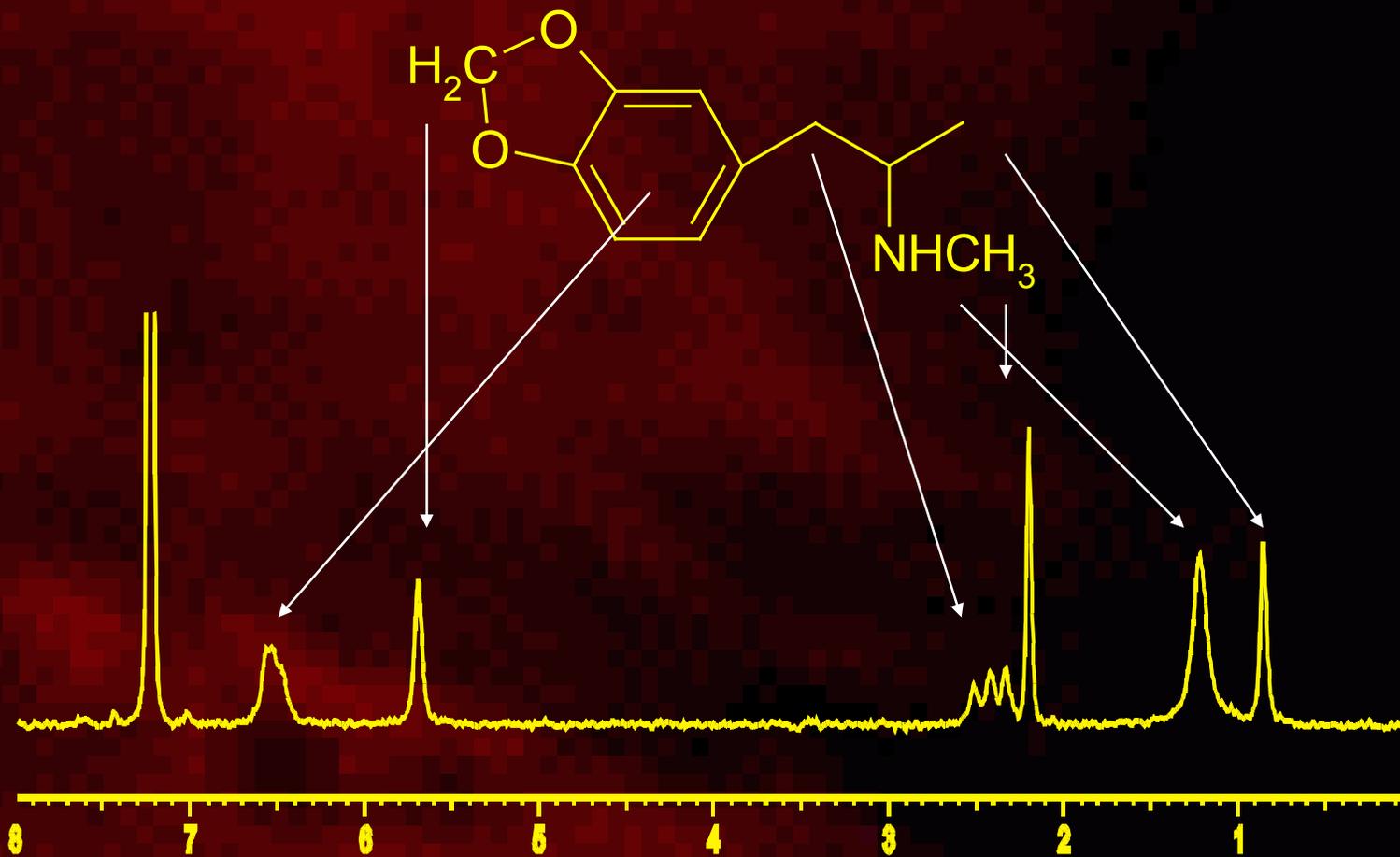
- ▽ RN/883/00
- △ RN/6108/00
- + RN/2932/99
- × RN/1491/00
- RN/1061/00



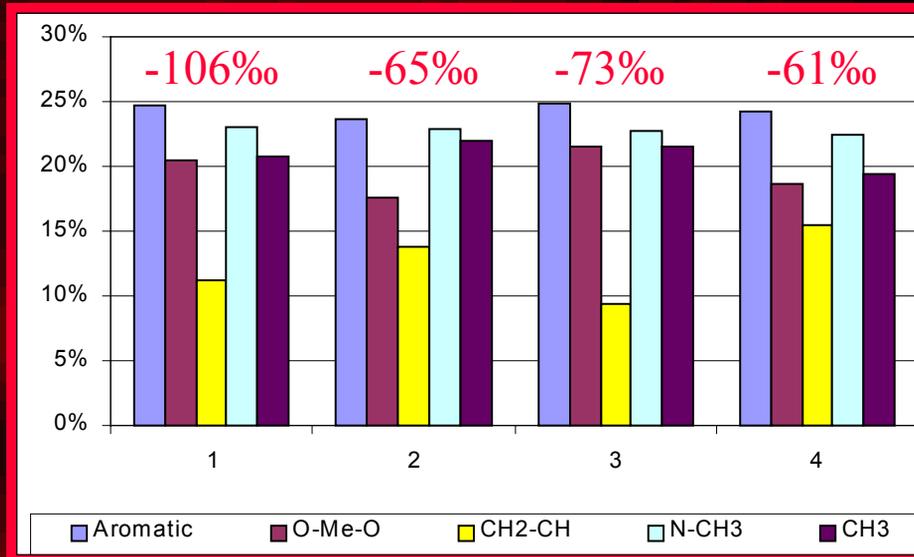
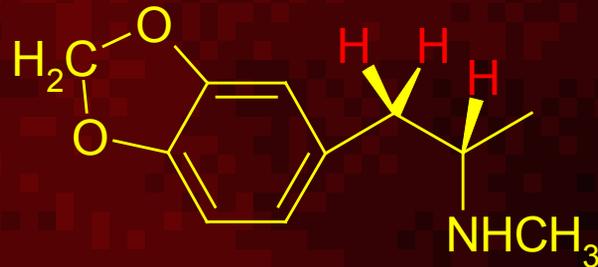
# $^2\text{H}$ SNIF-NMR

- Stable Natural Isotope Fractionation - NMR
- Site specific  $^2\text{H}$  substitution
- Jeol Alpha 500MHz
- 500-1000mg analyte in  $\text{CHCl}_3$
- 30,000 scans = 22 hours

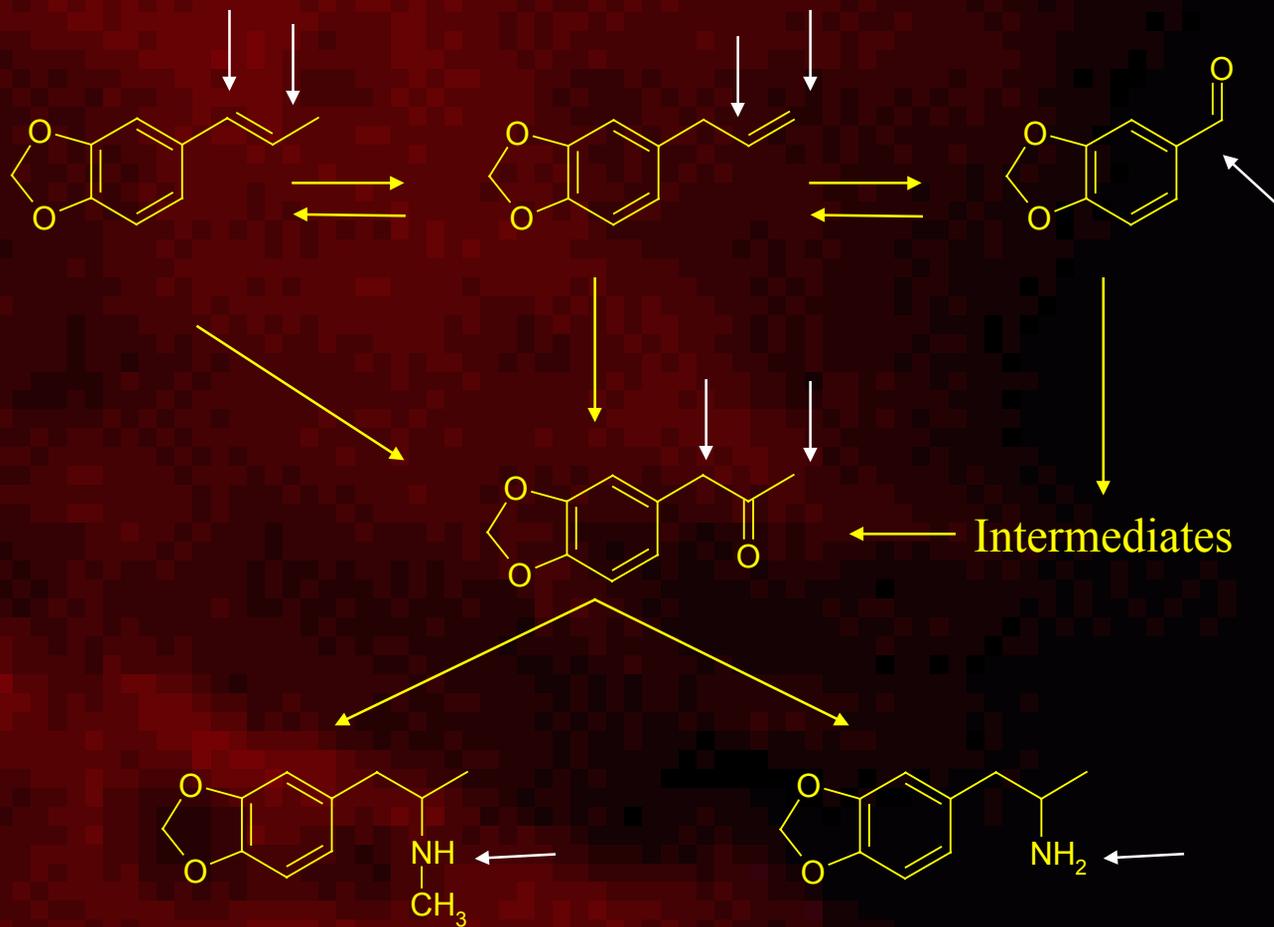
# $^2\text{H}$ SNIF-NMR



# $^2\text{H}$ SNIF-NMR



# Synthesis

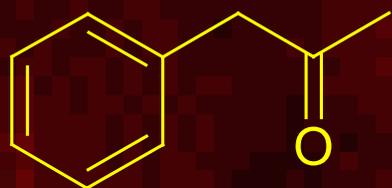


# Study II



$$Ca = (9Ck + Cm) / 10$$

# $\delta^{13}\text{C}$ results



$-29.25\text{‰} \pm 0.37$

+



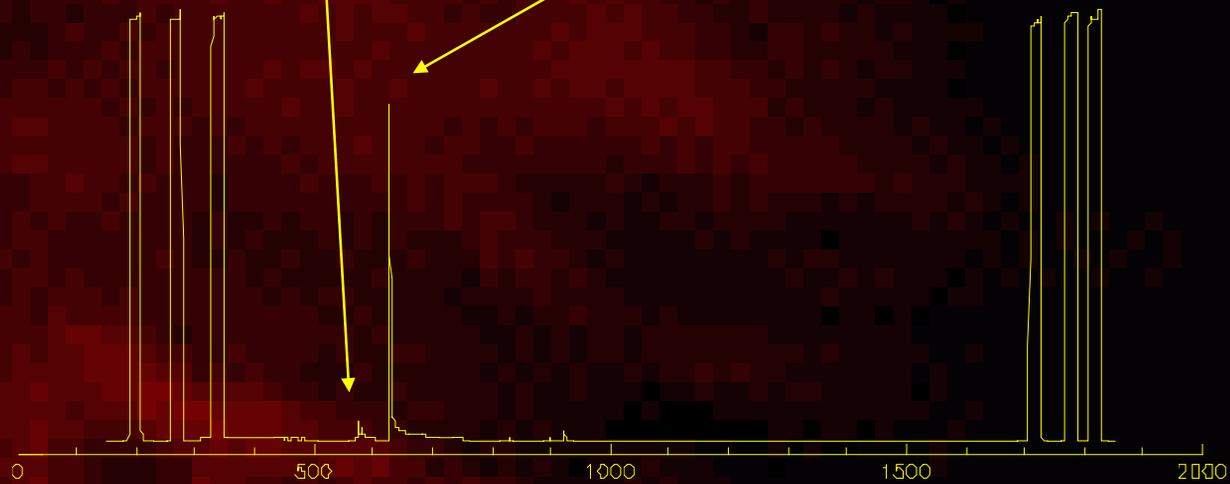
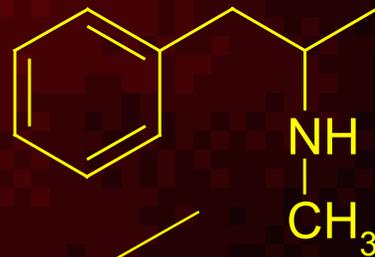
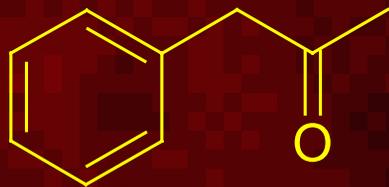
$-30.76\text{‰} \pm 0.12$



$-29.4\text{‰} \pm 0.39$

$-29.01\text{‰} \pm 0.02$

$-29.34\text{‰} \pm 0.57$



# $\delta^{13}\text{C}$ results



$-29.25\text{‰} \pm 0.37$

$-30.76\text{‰} \pm 0.12$

$-29.4\text{‰} \pm 0.39$

$-26.02\text{‰} \pm 0.18$

$-23.47\text{‰} \pm 0.26$

# $\delta^{15}\text{N}$ results



$-1.54\text{‰} \pm 0.16$

$+7.73\text{‰} \pm 0.05$

$+4.20\text{‰} \pm 0.51$

# $\delta^{15}\text{N}$ results



$-1.54\text{‰} \pm 0.16$

$+0.90\text{‰} \pm 0.63$

$+7.73\text{‰} \pm 0.05$

$+4.20\text{‰} \pm 0.51$

$+7.24\text{‰} \pm 0.11$

$+0.56\text{‰} \pm 0.06$

# Conclusion

- Irm-MS provides a “fingerprint” or “DNA”
- $\delta^{15}\text{N}$  major discriminating factor
- $\delta^2\text{H}$  and  $\delta^{13}\text{C}$  minor factors
- $\delta^{13}\text{C}$  and  $\delta^{15}\text{N}$  reflect reductive amination
- $\delta^2\text{H}$  reflects origin and “solvent” history
- Applicable to few 100ng material
- Irm-MS and  $^2\text{H}$ -NMR reveal synthesis

# Acknowledgments

- NERC Organic Mass Spectrometry Facility
- Andy Stott and Helen Grant  
NERC  $^{15}\text{N}$  Stable Isotope Facility
- Hugh Grundy & Eve Mason  
Avon and Somerset Constabulary  
Scientific Investigations
- Arndt Schimmelman  
Indiana University
- Prof Tim Gallagher